Mechanical Properties and Oxidation Behavior of Multilayered Hf–Si–N Coatings

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Abstract: Monolithic Hf–Si–N coatings and multilayered Hf–Si–N coatings with cyclical gradient concentration were fabricated using reactive direct current magnetron cosputtering. The structure of the Hf–Si–N coatings varied from a crystalline HfN phase, to a mixture of HfN and amorphous phases and to an amorphous phase with continuously increasing the Si content. The multilayered Hf48Si3N49 coatings exhibited a mixture of face-centered cubic and near-amorphous phases with a maximal hardness of 22.5 GPa, a Young’s modulus of 244 GPa and a residual stress of −1.5 GPa. The crystalline phase-dominant coatings exhibited a linear relationship between the hardness and compressive residual stress, whereas the amorphous phase-dominant coatings exhibited a low hardness level of 15–16 GPa; this hardness is close to that of Si3N4. Various oxides were formed after annealing of the Hf–Si–N coatings at 600 °C in a 1% O2–99% Ar atmosphere. Monoclinic HfO2 formed after Hf54N46 annealing and amorphous oxide formed for the oxidation-resistant Hf32Si19N49 coatings. The oxidation behavior with respect to the Si content was investigated by using transmission electron microscopy and X-ray photoelectron spectroscopy.

Keywords: mechanical properties; multilayered coatings; oxidation; transmission electron microscopy; X-ray photoelectron spectroscopy

1. Introduction

The group IVB (Ti, Zr and Hf) metal nitride coatings possess extraordinary characteristics of hardness [1–4], corrosion resistance [5–7] and decoration [8], but demonstrate inadequate oxidation resistance [9,10]. Ti–Si–N [11,12] and Zr–Si–N [13,14] have displayed enhanced oxidation resistance by Si addition. Although Hf–Si–N coatings have been applied as gate electrodes [15,16], corrosion-resistant films [17] and biocompatible films [18], few studies have focused on improving their oxidation resistance. In our previous study [19], Zr–Si–N coatings of 15–30 at.% Si maintained near-amorphous phases as examined through X-ray diffraction and exhibited excellent oxidation resistance at 600 °C in a 1% O2–99% Ar atmosphere for up to 100 h of annealing. These Zr–Si–N coatings with a cyclical gradient concentration fabricated using cosputtering with a low substrate holder rotation speed formed a multilayered structure consisting of alternatively-stacked high- and low-Si-content sublayers, which exhibited the characteristics of high oxidation resistance and mechanical properties, respectively [19]. The significant oxidation resistance was attributed to the lack of oxygen diffusion paths in near-amorphous phases without evident grain boundaries [20]. In this study, the effects of Si-addition on the mechanical properties and oxidation resistance of Hf–Si–N coatings periodically...
stacked of various Si-content sublayers were investigated. The phase and bonding characteristics’ variations after annealing were examined.

2. Materials and Methods

Hf–Si–N coatings were prepared on Si substrates at room temperature through reactive direct current magnetron cosputtering. Figure 1a shows the schematic of the cosputtering equipment (MGS-500, JUNSUN, New Taipei, Taiwan). Monolithic and multilayered Hf–Si–N coatings were prepared at substrate holder rotation speeds of 0 and 5 rpm, respectively. The monolithic coatings were deposited with a Ti interlayer. The Ti interlayer was deposited using an Ar flow of 20 sccm in a working pressure of 0.4 Pa, a substrate holder rotation speed of 5 rpm and a DC power of 200 W for 9 min, which formed an interlayer of approximately 100 nm. The target-to-substrate distance was kept at 90 mm for all sputtering runs. The flow rates of Ar and N2 gas were 12 and 8 sccm for fabricating nitride coatings in a working pressure of 0.4 Pa, respectively. The sample positions are labeled in Figure 1b. Targets of 99.95% pure Ti, 99.95% Hf and 99.999% Si with diameters of 50.8 mm were adopted as source materials. The sputtering equipment and cosputtering processes for fabricating Hf–Si–N coatings were similar to those for preparing Ta–Si–N and Zr–Si–N coatings, described in detail in previous studies [19,21]. The annealing environment was 1% O2–99% Ar at 600 °C; this condition was an oxidation-promoting atmosphere for evaluating the performance of protective coatings on glass molding dies [19,21].

Chemical composition analysis of the samples was carried out with a field-emission electron probe microanalyzer (FE-EPMA, JXA-8500F, JEOL, Akishima, Japan). The thickness of coatings was evaluated by field emission scanning electron microscopy (FE-SEM, S4800, Hitachi, Tokyo, Japan). An X-ray diffractometer (XRD, X’Pert PRO MPD, PANalytical, Almelo, The Netherlands) with Cu Kα radiation was used to analyze the phases of the coatings, using the grazing incidence technique with an incidence angle of 1°. The hardness (H) and elastic modulus (E) values of coatings were measured using a nanoindentation tester (TI-900 Triboindenter, Hysitron, Minneapolis, MN, USA) equipped with a Berkovich diamond probe tip. The indentation depth was 80 nm. The H and E values were calculated based on the Oliver and Pharr method [22]. The residual stress of the films was calculated using Stoney’s equation [23]:

\[ \sigma_{t} = \frac{E_S h_S^2}{6(1 - \nu_S) R_f} \]

where \( \sigma_t \) is the in-plane stress component in the film, \( t_f \) is the thickness of the film, \( E_S \) is the Young’s modulus of the Si substrate (130.2 GPa), \( \nu_S \) is the Poisson's ratio for the Si substrate (0.279) [24], \( h_S \) is the thickness of the substrate (525 μm) and \( R_f \) is the radius of the curvature of the film. The nanostructure of the annealed coatings was examined using transmission electron microscopy (TEM, JEM-2010F, JEOL, Tokyo, Japan). TEM samples were prepared by applying a focused ion beam system (FEI Nova 200,
Hillsboro, OR, USA). A Pt layer was deposited to protect the free surface during sample preparation. The chemical states of the constituent elements were examined by using an X-ray photoelectron spectroscopy (XPS, PHI 1600, PHI, Kanagawa, Japan) with an Mg Kα X-ray beam (energy = 1253.6 eV and power = 250 W) operated at 15 kV. The calibration of the binding energy scale was done with the C 1s line (284.6 eV) from the carbon contamination layer. An Ar⁺ ion beam of 3 keV was used to sputter the coatings for depth profiling. The sputter etching rate was set at 9.5 or 25.6 nm/min for SiO₂.

3. Results

3.1. As-Deposited Hf–Si–N Coatings

3.1.1. Monolithic Hf–Si–N Coatings

Table 1 shows the chemical compositions and coating thicknesses of six monolithic samples prepared using a \( W_{Hf} \) of 250 W and a \( W_{Si} \) of 25 W for 85 min while the substrate holder was not rotated. The positions of these coatings on the substrate holder are labeled in Figure 1b. The oxygen content in the as-deposited states was 0.6–1.9 at.%. The samples were designated in the form Hf\_x\_Si\_y\_N\_{100-x-y}.

Figure 2 illustrates the XRD patterns of the six monolithic Hf–Si–N coatings. The samples with 1 at.% Si at Positions 2 and 3 exhibited face-centered cubic (f.c.c.) HfN (111) and (200) reflections [ICDD 00-033-0592] with evident peak shifting toward lower 2θ values. The phase diagram of Hf–N binary alloy depicted equilibrium phases of HfN, Hf₃N₂ and Hf₄N₃ [25]; moreover, a metastable Hf₃N₄ was reported [16,26–28]. The two-theta angles of the main reflections of Hf₃N₂ [ICDD 01-073-1114] and Hf₄N₃ [ICDD 01-073-1115] were 35.81° and 35.295°, respectively, which were further away from the observed reflections. The sample with 3 at.% Si at Position 4 exhibited a mixture of HfN and Ti phases. The detected Ti reflections for the samples at Positions 4–6 were due to a low coating thickness level of 628–667 nm. The samples with 6–7 at.% Si at Positions 1 and 5 exhibited a mixture of HfN, Ti and near-amorphous phases, and the dominant reflection of the f.c.c. phase was HfN (200). The sample with 12 at.% Si at Position 6 had an X-ray amorphous phase-dominant structure accompanied by HfN and Ti reflections. Table 1 presents the mechanical properties and residual stress values of the monolithic Hf–Si–N coatings. The hardness of the monolithic Hf–Si–N coatings decreased with increasing Si content. The monolithic Hf–Si–N coatings with 1–3 at.% Si exhibited high mechanical properties (21.3–21.8 GPa) and compressive residual stress (−1.6 to −2.0 GPa) accompanied by evident HfN (111) reflections (Figure 2).

<table>
<thead>
<tr>
<th>Position</th>
<th>Sample</th>
<th>Chemical Composition (at.%)</th>
<th>T (nm)</th>
<th>H (GPa)</th>
<th>E (GPa)</th>
<th>Stress (GPa)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Hf₄Si₄N₄₀</td>
<td>44.5 ± 0.9</td>
<td>49.0 ± 1.6</td>
<td>1.0 ± 0.1</td>
<td>1009</td>
<td>19.5 ± 0.4</td>
</tr>
<tr>
<td>2</td>
<td>Hf₄Si₄N₃₀</td>
<td>48.3 ± 1.8</td>
<td>49.6 ± 2.1</td>
<td>0.9 ± 0.4</td>
<td>1359</td>
<td>21.7 ± 1.1</td>
</tr>
<tr>
<td>3</td>
<td>Hf₃Si₃N₄₀</td>
<td>52.1 ± 0.8</td>
<td>45.5 ± 0.8</td>
<td>1.1 ± 0.3</td>
<td>1122</td>
<td>21.3 ± 1.4</td>
</tr>
<tr>
<td>4</td>
<td>Hf₃Si₃N₃₇</td>
<td>50.0 ± 1.4</td>
<td>45.9 ± 1.2</td>
<td>1.3 ± 0.3</td>
<td>667</td>
<td>21.8 ± 1.3</td>
</tr>
<tr>
<td>5</td>
<td>Hf₃Si₃N₃₉</td>
<td>43.4 ± 0.3</td>
<td>48.1 ± 0.7</td>
<td>1.9 ± 0.5</td>
<td>628</td>
<td>19.3 ± 2.2</td>
</tr>
<tr>
<td>6</td>
<td>Hf₃Si₂N₄₀</td>
<td>38.1 ± 0.8</td>
<td>49.4 ± 0.5</td>
<td>0.6 ± 0.2</td>
<td>649</td>
<td>15.4 ± 1.1</td>
</tr>
</tbody>
</table>

Note: \( T \) stands for thickness; \( H \) stands for hardness; \( E \) stands for Young’s modulus.
3.1.2. Multilayered Hf–Si–N Coatings

The multilayered coatings were constructed by cyclical gradient concentration deposition as the substrate holder rotation speed was maintained at a low level [19,21]. According to the calculation model for multilayered Zr–Si–N [19] and Ta–Si–N [21] based on the chemical compositions and thicknesses of the monolithic coatings:

$$\langle C(X) \rangle = \frac{\sum C(X) \Delta \theta \Delta t}{\sum \Delta \theta \Delta t}$$  \hspace{1cm} (2)

where $\Delta \theta$ is the angle between the two adjacent sample positions and $\Delta t$ and $C(X)$ are the average coating thickness and element $X$ content of the adjacent two samples prepared without rotating the substrate-holder, respectively, the chemical composition of a Hf–Si–N coating prepared using sputtering powers of $W_{\text{Hf}} = 250$ W and $W_{\text{Si}} = 25$ W and a substrate holder rotation speed of 5 rpm was determined to be Hf$_{48}$Si$_{32}$N$_{49}$. This composition was close to the real composition, Hf$_{45}$Si$_{33}$N$_{49}$, as listed in Table 2. Table 2 shows the chemical compositions of the as-deposited multilayered HfN and Hf–Si–N coatings prepared at various sputtering powers and a substrate holder rotation speed of 5 rpm. The oxygen content in the as-deposited states was 0.9–3.2 at.%. Hf$_{54}$N$_{46}$ coatings exhibited a thickness of 957 nm after sputtering for 90 min, representing a deposition rate of 10.6 nm/min. At a fixed Hf sputtering power ($W_{\text{Hf}}$) of 250 W, the deposition rates of the Hf–Si–N coatings were 10.2, 12.7, 12.8, 14.1 and 16.5 nm/min at Si target powers ($W_{\text{Si}}$) of 25, 50, 75, 100 and 150 W and sputter times of 85, 80, 75, 70 and 60 min, respectively; the coating thicknesses were in the range of 866–1018 nm.

**Table 2.** Sputtering parameters, chemical compositions, thicknesses and mechanical properties of multilayered Hf–Si–N coatings.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Power (W)</th>
<th>Time (min)</th>
<th>Chemical Composition (at.%)</th>
<th>Thickness (nm)</th>
<th>Hardness (GPa)</th>
<th>Young's Modulus (GPa)</th>
<th>Stress (GPa)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hf$<em>{54}$N$</em>{46}$</td>
<td>250</td>
<td>0</td>
<td>90</td>
<td>52.2 ± 4.0</td>
<td>0.2 ± 0.1</td>
<td>44.4 ± 4.5</td>
<td>957</td>
</tr>
<tr>
<td>Hf$<em>{54}$Si$</em>{25}$N$_{49}$</td>
<td>250</td>
<td>25</td>
<td>85</td>
<td>47.2 ± 0.7</td>
<td>3.1 ± 0.1</td>
<td>47.8 ± 0.3</td>
<td>866</td>
</tr>
<tr>
<td>Hf$<em>{54}$Si$</em>{50}$N$_{46}$</td>
<td>250</td>
<td>50</td>
<td>80</td>
<td>44.7 ± 0.5</td>
<td>7.0 ± 0.2</td>
<td>47.1 ± 0.1</td>
<td>1018</td>
</tr>
<tr>
<td>Hf$<em>{54}$Si$</em>{75}$N$_{49}$</td>
<td>250</td>
<td>75</td>
<td>75</td>
<td>39.0 ± 0.8</td>
<td>11.4 ± 0.4</td>
<td>48.5 ± 1.4</td>
<td>961</td>
</tr>
<tr>
<td>Hf$<em>{54}$Si$</em>{100}$N$_{51}$</td>
<td>250</td>
<td>100</td>
<td>70</td>
<td>35.4 ± 1.0</td>
<td>13.4 ± 0.1</td>
<td>50.3 ± 1.2</td>
<td>985</td>
</tr>
<tr>
<td>Hf$<em>{54}$Si$</em>{150}$N$_{49}$</td>
<td>250</td>
<td>150</td>
<td>60</td>
<td>31.8 ± 0.4</td>
<td>18.5 ± 0.5</td>
<td>48.7 ± 1.0</td>
<td>990</td>
</tr>
</tbody>
</table>

Figure 3 illustrates the XRD patterns of the as-deposited multilayered Hf–Si–N coatings. The XRD pattern of the Hf$_{54}$N$_{46}$ coatings revealed an f.c.c. HfN phase. Hf$_{48}$Si$_{32}$N$_{49}$ coatings exhibited a mixture of...
f.c.c. and near-amorphous phases, whereas Hf$_{46}$Si$_7$N$_{48}$ coatings were dominated by a near-amorphous phase accompanied with a minor f.c.c. (200) reflection, and the (111) reflection could overlap a broadened amorphous reflection. The f.c.c. reflections of the Hf$_{44}$N$_{46}$ and Hf$_{48}$Si$_3$N$_{49}$ coatings shifted to lower 20 values, implying an expanded lattice. The residual stresses of the crystalline Hf$_{54}$N$_{46}$ and Hf$_{48}$Si$_3$N$_{49}$ coatings were $-2.4$ and $-1.5$ GPa (Table 2), respectively. The Si atoms dissolved as interstitial atoms in the HfN structure, which expanded the lattice parameters of an f.c.c. HfN phase, resulting a residual stress in compression. The Hf$_{46}$Si$_7$N$_{48}$ coatings exhibited a residual stress of $-0.7$ GPa. The Hf–Si–N coatings with a Si content in the range of 12–19 at.% exhibited a near-amorphous phase and near-zero residual stress that ranged from $-0.4$ to $-0.5$ GPa.

Figure 3 illustrates the XRD patterns of the as-deposited multilayered Hf–Si–N coatings. Figure 4 illustrates the nanoindentation hardness and Young’s modulus of the as-deposited multilayered Hf–N and Hf–Si–N coatings, which exhibit similar trends as varying the Si content. The hardness of the as-deposited Hf$_{54}$N$_{46}$ coatings was $22.3 \pm 0.6$ GPa, which is comparable to the reported values of 19–21 GPa [4,29,30] for HfN films prepared through sputtering. Seo et al. [3] reported a high hardness value of 25.2 GPa for epitaxial HfN(001) layers. As Si was introduced successively into the coatings, the hardness varied: the Hf$_{48}$Si$_3$N$_{49}$ coatings had a hardness of $22.5 \pm 0.8$ GPa, whereas the hardness decreased sharply to $15.3 \pm 0.6$ GPa for the Hf$_{46}$Si$_7$N$_{48}$ coatings and then remained at $15$–$16$ GPa for the Hf$_{39}$Si$_{12}$N$_{49}$, Hf$_{36}$Si$_{13}$N$_{51}$ and Hf$_{32}$Si$_{19}$N$_{49}$ coatings; this level is close to $17$–$19$ GPa for Si$_3$N$_4$ [31–34].

Figure 4. Nanoindentation hardness and Young’s modulus values of the as-deposited multilayered Hf–Si–N coatings.
Figure 5 shows the relationship of hardness and residual stress of the monolithic and multilayered Hf-Si-N coatings. Two categories of the Hf-Si-N coatings were observed. The crystalline phase-dominant coatings exhibited a linear relationship between hardness and residual stress, whereas the X-amorphous phase-dominant coatings maintained a relatively low residual stress level accompanied by a low hardness level of 15–16 GPa.

![Graph](image)

**Figure 5.** Nanoindentation hardness and residual stress values of the monolithic and multilayered Hf-Si-N coatings.

The bonding characteristics of three typical Hf-Si-N coatings, a crystalline and amorphous mixed Hf$_{48}$Si$_3$N$_{49}$, an X-ray amorphous dominated Hf$_{48}$Si$_7$N$_{48}$ and an X-ray amorphous Hf$_{32}$Si$_{19}$N$_{49}$ coatings, respectively representing low-, medium- and high-Si-contents Hf-Si-N coatings, were analyzed. Figure 6 shows the XPS depth profiles of the N 1s, O 1s, Si 2p and Hf 4f core levels of the Hf$_{48}$Si$_3$N$_{49}$ coatings. The binding energies of N 1s were identified as 396.88 ± 0.02 eV at a depth range of 9.5–57 nm, labeled as sputter times of 1–6 min in Figure 6a, which was comparable to the reported values of 396.9 eV for the HfN and Hf$_3$N$_2$/Hf$_4$N$_3$ compounds [35]. Because the binding energies between various N–Hf bonds were not distinguishable [36], the characteristics of Hf–N bonds were verified by the Hf signals. The O 1s signal was observed on the free surface (not shown in Figure 6b), whereas beneath the surface, the profile at approximately 534 eV was overlapped with an Hf 4s signal. No Si signal was detected beneath the free surface due to a low-Si-content of 3 at.%. Arranz [35] reported that the binding energies of Hf 4f$_{7/2}$ were identified as 13.9, 14.5, 15.0 and 15.5 eV for the Hf$_6$, Hf$_3$N$_2$/Hf$_4$N$_3$, HfN and Hf$_3$N$_4$ components, respectively. The Hf–N bonds for Hf$_3$N$_2$ and Hf$_4$N$_3$ were not distinct and denoted using Hf$_4$N$_3$. 

![Graphs](image)

**Figure 6.** Cont.
The presence of metallic Hf and decreased amounts of HfN and Hf3N4 bonds related to those of the Hf48Si3N49 coatings implied the introduced Si preferential bonding with N.
annealed, 24 h

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coatings [39]. Zr coatings implied that the amount of Si–N bonds was saturated and Hf tended to form the Hf$_3$N$_4$ and the presence of free Si and increased amounts of Hf

of 98.40 ± 0.06 and 100.87 ± 0.04 eV for the free Si (un-nitrified Si [37]) and Si–N bonds [38] in the interior region, respectively. The Hf 4f signals consisted of Hf$_4$N$_3$, HfN, Hf$_3$N$_4$ and Hf–O (Figure 10c), the intensity ratio of which was 23:26:40:11 for Hf$_4$N$_3$:HfN:Hf$_3$N$_4$:Hf–O. The absence of metallic Hf$^0$ and the presence of free Si and increased amounts of Hf$_3$N$_4$ bonds related to those of the Hf$_{48}$Si$_{19}$N$_{49}$ coatings implied that the amount of Si–N bonds was saturated and Hf tended to form the Hf$_3$N$_4$ compound accompanied by a high Si content, which was similar to that reported for the Zr–Si–N coatings [39]. Zr$_3$N$_4$ and Hf$_3$N$_4$ with the Th$_3$P$_4$ structure have been reported [27,28,40].

![Figure 8. Curve fitting of the Hf 4f signal of Hf$_{48}$Si$_{19}$N$_{49}$ coatings at a depth of 57 nm.](image)

Figure 9 shows the XPS depth profiles of the N 1s, O 1s, Si 2p and Hf 4f core levels of the Hf$_{32}$Si$_{19}$N$_{49}$ coatings. The binding energies in the shallow region, a depth range of 9.5–19.0 nm, deviated from those in an interior region of 28.5–57.0 nm. Figure 10 shows the curve fitting of the N 1s, Si 2p and Hf 4f signal at a depth of 57 nm. In the interior region, the N 1s of Hf$_{32}$Si$_{19}$N$_{49}$ coatings comprised two signals of 396.49 ± 0.06 and 397.17 ± 0.06 eV (Figure 10a), which respectively fitted the binding energies of N–Hf and N–Si bonds. The Si 2p (Figure 10b) comprised two signals of 98.40 ± 0.06 and 100.87 ± 0.04 eV for the free Si (un-nitrified Si [37]) and Si–N bonds [38] in the interior region, respectively. The Hf 4f signals consisted of Hf$_4$N$_3$, HfN, Hf$_3$N$_4$ and Hf–O (Figure 10c), the intensity ratio of which was 23:26:40:11 for Hf$_4$N$_3$:HfN:Hf$_3$N$_4$:Hf–O. The absence of metallic Hf$^0$ and the presence of free Si and increased amounts of Hf$_3$N$_4$ bonds related to those of the Hf$_{48}$Si$_{19}$N$_{49}$ coatings implied that the amount of Si–N bonds was saturated and Hf tended to form the Hf$_3$N$_4$ compound accompanied by a high Si content, which was similar to that reported for the Zr–Si–N coatings [39]. Zr$_3$N$_4$ and Hf$_3$N$_4$ with the Th$_3$P$_4$ structure have been reported [27,28,40].

### Table 3. XPS analysis results of Hf 4f$_{7/2}$ of as-deposited and annealed Hf–Si–N coatings.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Depth Range (nm)</th>
<th>Hf 4f$_{7/2}$ (eV)</th>
<th>Hf–O</th>
<th>Intensity Ratio of Hf$^0$:Hf$_4$N$_3$:HfN:Hf$_3$N$_4$:Hf–O</th>
</tr>
</thead>
<tbody>
<tr>
<td>as-deposited</td>
<td>9.5–57.0</td>
<td>14.46 ± 0.06</td>
<td>15.13 ± 0.03</td>
<td>16.01 ± 0.07 18.11 ± 0.09 0:31:24:34:11</td>
</tr>
<tr>
<td>annealed, 24 h</td>
<td>205–256, unoxidized</td>
<td>13.84 ± 0.02</td>
<td>14.48 ± 0.01</td>
<td>15.02 ± 0.02 16.01 ± 0.00 17.50 ± 0.07 18:29:34:12:7</td>
</tr>
<tr>
<td>Hf$<em>{48}$Si$</em>{19}$N$_{49}$</td>
<td>26–179, oxidized</td>
<td>13.85 ± 0.02</td>
<td>14.48 ± 0.03</td>
<td>14.98 ± 0.01 16.07 ± 0.04 17.51 ± 0.02 11:2:5:30:52</td>
</tr>
<tr>
<td>as-deposited</td>
<td>9.5–57.0</td>
<td>14.53 ± 0.01</td>
<td>14.82 ± 0.03</td>
<td>15.82 ± 0.03 17.06 ± 0.08 28:36:11:12:13</td>
</tr>
<tr>
<td>annealed, 8 h</td>
<td>161–230, unoxidized</td>
<td>13.86 ± 0.01</td>
<td>14.50 ± 0.02</td>
<td>15.00 ± 0.28 15.72 ± 0.09 16.90 ± 0.03 23:39:10:16:12</td>
</tr>
<tr>
<td>Hf$<em>{32}$Si$</em>{19}$N$_{49}$</td>
<td>23–115, oxidized</td>
<td>13.50 ± 0.02</td>
<td>14.84 ± 0.08</td>
<td>15.09 ± 0.07 16.21 ± 0.04 16.97 ± 0.02 5:9:15:10:61</td>
</tr>
<tr>
<td>as-deposited</td>
<td>28.5–57.0</td>
<td>14.60 ± 0.02</td>
<td>15.25 ± 0.01</td>
<td>16.07 ± 0.03 17.53 ± 0.06 0:23:26:40:11</td>
</tr>
<tr>
<td>annealed, 24 h</td>
<td>16.8–33.6, unoxidized</td>
<td>13.92 ± 0.02</td>
<td>14.53 ± 0.03</td>
<td>15.18 ± 0.02 15.70 ± 0.03 16.98 ± 0.05 18:29:30:16:7</td>
</tr>
<tr>
<td>8.4, oxidized</td>
<td>14.60 ± 0.02</td>
<td>15.25 ± 0.01</td>
<td>16.07 ± 0.03</td>
<td>17.53 ± 0.06 0:23:26:40:11</td>
</tr>
</tbody>
</table>
Moreover, cross-sectional SEM images showed lateral cracks in the oxide scale of the Hf–Si–N coatings implied that the introduced Si preferentially bonded with N.

Figure 9. XPS depth profiles of the Hf$_{32}$Si$_{19}$N$_{49}$ coatings: (a) N 1s, (b) O 1s, (c) Si 2p, (d) Hf 4f (sputter rate: 9.5 nm/min for SiO$_2$).

Figure 10. Curve fitting of XPS profiles of: (a) N 1s, (b) Si 2p and (c) Hf 4f signals of the Hf$_{32}$Si$_{19}$N$_{49}$ coatings at a depth of 57 nm.

The oxide depths of the annealed Hf–Si–N coatings examined using SEM. The oxide depths of the annealed Hf–Si–N coatings implied the introduced Si preferentially bonding with N.
3.2. Annealed Hf–Si–N Coatings

Figure 11 illustrates the XRD patterns of the multilayered Hf–Si–N coatings after annealing at 600 °C in 1% O₂–99% Ar for 4 h. A monoclinic HfO₂ phase [ICDD 34-0104] was observed for the annealed Hf₅₄N₄₆ and Hf₄₈Si₃N₄₉ coatings in addition to the original f.c.c. phase, with reflections shifted back to the standard values. By contrast, the Hf–Si–N coatings with Si contents of 7–19 at.% exhibited phases similar to those of the as-deposited states. Figure 12 illustrates the oxide scale thicknesses of the annealed Hf–Si–N coatings examined using SEM. The oxide depths of the annealed Hf₅₄N₄₆ coatings were too low to be examined using SEM. The Hf₅₄N₄₆ coatings exhibited a rapid increase in the oxide scale thickness as the annealing duration was increased from 12 to 24 h. Moreover, cross-sectional SEM images showed lateral cracks in the oxide scale of the Hf₅₄N₄₆ coatings after annealing for 24–100 h (Figure 13). The volumes of HfN and HfO₂ were 0.02309 and 0.03457 nm³/metal atom, respectively. Therefore, the volume ratio of oxide/nitride was 1.50, which resulted in cracks forming.

Figure 11. XRD patterns of multilayered Hf–Si–N coatings after they were annealed at 600 °C in 1% O₂–99% Ar for 4 h.

Figure 12. Oxide scale thicknesses of multilayered Hf–Si–N coatings after they were annealed at 600 °C in 1% O₂–99% Ar.
The oxidation scale thicknesses of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings examined using SEM were 146, 175 and 180 nm after annealing for 24, 50 and 100 h, respectively (Figure 12), which exhibited a decreasing oxidation rate. The XRD pattern of the 8-h annealed Hf$_{48}$Si$_{3}$N$_{49}$ coatings exhibited a mixture of monoclinic and cubic HfO$_2$ [ICDD 00-053-0550] and HfN phases, which was maintained until 100 h of annealing (Figure 14). Figure 15a illustrates a cross-sectional TEM image of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings after annealing for 4 h at 600 °C in 1% O$_2$–99% Ar. An oxide layer was observed on the surface, and the selected area diffraction pattern (SADP) from the near-surface region exhibited ring patterns of HfN and monoclinic HfO$_2$ phases. Figure 15b,c exhibits dark-filed images correlated with the HfO$_2$(–111) and HfN(200) diffraction spots of the SADP, respectively indicating the surface oxide layer and interior columnar structure. The high-resolution image of the oxide scale exhibited lattice fringes of monoclinic HfO$_2$ (Figure 15d). Figure 16a illustrates a cross-sectional TEM image of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings after annealing for 24 h. The oxide scale was divided into two sublayers. The near-surface region labeled as Oxide Scale I was an amorphous structure, whereas Oxide Scale II at a higher depth was crystalline. The high-resolution image of Oxide Scale II exhibited lattice fringes of monoclinic and cubic HfO$_2$ (Figure 16b), whereas the high-resolution image of the unoxidized region showed lattice fringes of HfN (Figure 16c). Figure 17a illustrates a cross-sectional TEM image of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings after annealing for 100 h; the interface between the two sublayers observed for the 24-h annealed sample was not clear. The high-resolution image of the oxide scale near the surface exhibited lattice fringes of the monoclinic HfO$_2$ phase (Figure 17b), whereas the high-resolution image of the oxide scale at a higher depth showed lattice fringes of monoclinic and cubic HfO$_2$ (Figure 17c), similar to those in Oxide Scale II of the 24-h annealed sample. Because the standard Gibbs free energy levels of HfO$_2$ and SiO$_2$ at 600 °C are –983.438 and –752.535 kJ/mol of O$_2$ [41], respectively, HfO$_2$ formed preferentially during the oxidation of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings, as observed for the 4-h annealed sample. Subsequently, Si diffused outward in the following oxidation and formed an amorphous oxide scale on the free surface, which inhibited the inward diffusion of O; therefore, the oxidation of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings was restricted, as observed for the 24-h annealed sample. However, because the Si content of the Hf$_{48}$Si$_{3}$N$_{49}$ coatings was only 3 at.% and Hf became a sink for gettering O, oxidation proceeded continuously, and Oxide Scale I, amorphous Si-oxide, was exhausted and replaced by crystalline monoclinic HfO$_2$. 

Figure 13. Cross-sectional SEM image of the Hf$_{34}$N$_{46}$ coating after annealing at 600 °C in 1% O$_2$–99% Ar for 24 h.
Figure 14. XRD patterns of Hf$_{48}$Si$_3$N$_{49}$ coatings after they were annealed at 600 °C in 1% O$_2$–99% Ar for 0–100 h.

Figure 15. (a) Cross-sectional TEM image and selected area diffraction pattern of the Hf$_{48}$Si$_3$N$_{49}$ coating after annealing at 600 °C in 1% O$_2$–99% Ar for 4 h; the dark-field images correlated with diffraction spots of (b) HfO$_2$(−111) and (c) HfN(200) in the selected area diffraction pattern (SADP); (d) high-resolution image of the oxide layer.
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Figure 16. (a) Cross-sectional TEM image and selected area diffraction pattern of the Hf$_{48}$Si$_{3}$N$_{49}$ coating after annealing at 600 °C in 1% O$_2$–99% Ar for 100 h; high-resolution images of (b) Oxide Scale II and (c) the unoxidized region in (a).

Figure 17. (a) Cross-sectional TEM image of the Hf$_{48}$Si$_{3}$N$_{49}$ coating after annealing at 600 °C in 1% O$_2$–99% Ar for 24 h; high-resolution images of (b) Oxide Scale II and (c) the unoxidized region in (a).

Both the Hf$_{46}$Si$_{7}$N$_{48}$ and Hf$_{39}$Si$_{12}$N$_{49}$ coatings exhibited buckling after annealing for 12 h and detaching after annealing for 24 h, whereas the Hf$_{36}$Si$_{13}$N$_{51}$ and Hf$_{32}$Si$_{19}$N$_{49}$ coatings maintained
adherent to the substrates. Figure 18 illustrates the XRD patterns of the Hf–Si–N coatings after annealing for 100 h. The annealed Hf$_{52}$Si$_{19}$N$_{49}$ coatings exhibited a monoclinic HfO$_2$ phase, whereas the annealed Hf$_{36}$Si$_{13}$N$_{51}$ and Hf$_{52}$Si$_{19}$N$_{49}$ coatings retained near-amorphous phases. Figure 19a illustrates a cross-sectional TEM image of the Hf$_{52}$Si$_{19}$N$_{49}$ coatings after annealing at 600 °C in 1% O$_2$–99% Ar for 24 h. A shallow oxide scale less than 10 nm was observed on the free surface, and the image contrast of the coatings revealed a multilayered structure attributed to the cyclical gradient concentration deposition. The SADP from the near-surface region exhibited a near-amorphous phase. The high-resolution image of the oxide scale exhibited an amorphous phase, and lattice fringes of cubic HfO$_2$ were seldom observed in the Hf$_{52}$Si$_{19}$N$_{49}$ coatings (Figure 19b).

The high oxidation resistances of Hf$_{48}$Si$_{32}$N$_{49}$ and Hf$_{52}$Si$_{19}$N$_{49}$ coatings and reduced oxidation resistance of Hf$_{48}$Si$_7$N$_{48}$ coatings were further examined by XPS analyses. Figure 20 shows the XPS depth profiles of the 24-h annealed Hf$_{48}$Si$_7$N$_{48}$ coatings. The sputtering rate for XPS analysis was set at 25.6 nm/min. Two evident regions exhibited the oxidized and unoxidized parts of the annealed Hf$_{48}$Si$_7$N$_{48}$ coatings. The unoxidized region at a depth range of 205–256 nm exhibited Hf–N bonds whereas the oxidized region at a depth range of 205–256 nm exhibited Hf–O bonds with binding energies similar to those of the Hf–N bonds of the as-deposited samples (Table 3). Moreover, the intensity ratio of Hf$^4$:Hf$_2$N$_3$:HfN:Hf$_3$N$_4$:Hf–O changed from 0.31:2.43:1.1 to 18.29:34:12.7 after 600 °C annealing, which implied that Hf$_3$N$_4$ was not stable and decomposed to HfN and Hf$^4$. By contrast, the intensity ratio of the oxidized region at a depth range of 26–179 nm was 11.25:53:30:52, which consisted of major Hf–O bonds and Hf–N bonds of Hf$_3$N$_4$. The variation between
the Hf–N bonds indicated that Hf$_3$N$_3$ and HfN oxidized preferentially, but Hf$_5$N$_4$ maintained the original amounts.

Figure 20. XPS depth profiles of the Hf$_{48}$Si$_{32}$N$_{48}$ coatings after annealing for 24 h: (a) N 1s; (b) O 1s; (c) Si 2p; (d) Hf 4f. Sputter rate: 25.6 nm/min for SiO$_2$.

Table 3 lists the XPS Hf 4f$_{7/2}$ signals of the 24-h annealed Hf$_{32}$Si$_{19}$N$_{49}$ coatings. Because the oxide depth was less than 10 nm (Figure 19a), the sputtering rate of XPS analyses was set at 4.2 nm/min. The Hf 4f$_{7/2}$ signals in the unoxidized region at a depth of 16.8–33.6 nm comprised all the mentioned Hf$^0$, Hf–N and Hf–O bonds, and the intensity ratio of Hf$^0$:Hf$_3$N$_3$:HfN:Hf$_5$N$_4$:Hf–O was 18:29:30:16:7. The presence of metallic Hf$^0$ and declined intensity of Hf$_5$N$_4$ bonds related to those of the as-deposited Hf$_{32}$Si$_{19}$N$_{49}$ coatings implied that Hf$_3$N$_3$ was not stable as the coatings were annealed at 600 $^\circ$C, which agreed with the observation on the unoxidized region of the 24-h annealed Hf$_{48}$Si$_{32}$N$_{48}$ coatings. The Si 2p comprised two signals of 97.84 ± 0.11 and 100.25 ± 0.06 eV for the free Si and Si–N bonds, respectively. The oxidized region at a depth of 8.4 nm exhibited Hf 4f$_{7/2}$ signals of 16.00 and 16.78 eV, which were Hf–N bonds of Hf$_3$N$_3$ and Hf–O bonds, respectively, and the intensity ratio of Hf–N:Hf–O was 20:80.

The presence of metallic Hf$^0$ in the as-deposited state of the Hf$_{48}$Si$_7$N$_{48}$ coatings resulted in the highest oxidation rate among the Hf–Si–N coatings in the beginning stage of oxidation (Figure 12). The intensity ratio of various Hf species at the unoxidized region annealed for 8 h maintained a similar ratio, whereas the Hf$^0$ and Hf$_3$N$_3$ constitutions in the oxidized region diminished evidently.

Figures 21 and 22 illustrate the variations in residual stress and hardness values after annealing. The residual stress of the crystalline Hf$_5$N$_4$ coatings varied from compressive stress to zero stress during 50 h of annealing. By contrast, the nanoindentation hardness increased in the first 4 h of annealing and then decreased continuously until 24 h of annealing. Previous studies have reported hardness values of 9–12 GPa for HfO$_2$ phases [42–44]. The 12-h and 24-h annealed Hf$_5$N$_4$ coatings exhibited a hardness of 13.6 and 5.7 GPa, respectively. The residual stress of the annealed Hf$_{48}$Si$_{32}$N$_{48}$ coatings remained at −0.4 GPa after annealing for 4–24 h. The hardness decreased gradually to
The presence of metallic Hf0 in the as-deposited state of the Hf–Si–N coatings varied significantly. The low-Si-content (1–3 at.%) coatings exhibited an f.c.c. structure, whereas the middle-Si-content (6–12 at.%) coatings exhibited a mixture of HfN and amorphous phases. In the monolithic coatings, the low-Si-content (1–3 at.%) coatings exhibited an f.c.c. structure, whereas the middle-Si-content (6–12 at.%) coatings exhibited a mixture of HfN and amorphous phases. In the multilayered coatings, the Hf–Si–N coatings exhibited X-ray amorphous structures. The nanoindentation hardness of the as-deposited Hf–Si–N coatings varied from 13.6 to 5.7 GPa, respectively. The construction of multilayered coatings from monolithic sublayers was elucidated.

The residual stress of the annealed Hf–Si–N coatings varied from compressive stress to zero stress as the annealing time increased to 24 h and remained at this level for up to 100 h of annealing. Both the oxide scale and the unoxidized coating of the 24-h annealed Hf$_{48}$Si$_{33}$N$_{49}$ coating exhibited a dense structure without pores and cracks (Figure 16). All the near-amorphous Hf–Si–N coatings with 7–19 at.% Si remained in a stress-free state after 4 h of annealing. The 24-h annealed Hf$_{52}$Si$_{19}$N$_{49}$ coatings exhibited a high hardness level of 16.6 GPa and retained an amorphous structure for up to 100 h of annealing.

![Figure 21. Residual stress of multilayered Hf–Si–N coatings after they were annealed at 600 °C in 1% O$_2$–99% Ar.](image)

![Figure 22. Nanoindentation hardness values of the annealed multilayered Hf–Si–N coatings.](image)

4. Conclusions

Monolithic and multilayered Hf–Si–N coatings deposited on Si substrates were fabricated by using reactive direct current magnetron cosputtering with substrate holder rotation speeds of 0 and 5 rpm, respectively. The construction of multilayered coatings from monolithic sublayers was elucidated. In the monolithic coatings, the low-Si-content (1–3 at.%) coatings exhibited an f.c.c. structure, whereas the middle-Si-content (6–12 at.%) coatings exhibited a mixture of HfN and amorphous phases. In the multilayered coatings, the Hf$_{54}$N$_{46}$ coatings exhibited an f.c.c. structure, whereas the low-Si-content (3 at.%) coatings exhibited a mixture of f.c.c. and amorphous phases, and the high-Si-content (7–19 at.%) coatings exhibited X-ray amorphous structures. The nanoindentation hardness of the as-deposited multilayered Hf$_{48}$Si$_{33}$N$_{49}$ coatings exhibited a maximum value of 22.5 ± 0.8 GPa, and the hardness values decreased abruptly to 15.3 ± 0.6 GPa for Hf$_{46}$Si$_{48}$N$_{49}$ and remained at 15–16 GPa for the high-Si-content coatings, approaching that of Si$_3$N$_4$. Moreover, the crystalline phase-dominant
coatings of the surveyed monolithic and multilayered Hf–Si–N coatings exhibited a linear relationship between hardness and residual stress in compression. The XPS analyses indicated that Hf and Hf$_4$N$_3$ preferentially oxidized at 600 °C in an oxidizing circumstance, whereas Hf$_3$N$_4$ was not stable at 600 °C in non-oxidizing condition; therefore, the high oxidation resistances of Hf$_{48}$Si$_3$N$_{49}$ and Hf$_{32}$Si$_{19}$N$_{49}$ coatings were attributed to the performance of Si-addition in various mechanisms. The oxidation of Hf$_{48}$Si$_3$N$_{49}$ coatings was restricted because of the outward diffusion of Si, which formed an amorphous oxide scale on the free surface of the initially-formed HfO$_2$ oxide scale, although the amorphous oxide scale was replaced by HfO$_2$ after subsequent annealing. By contrast, the oxidation of the Hf$_{32}$Si$_{19}$N$_{49}$ coatings was restricted because their structures were maintained X-ray amorphous for up to 100 h of annealing.

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