

# Stable Anionic $\sigma$ -Complexes of Highly Electrophilic Aromatics and C-Nucleophiles: Synthesis and Oxidation <sup>†</sup>

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**Abstract:** Reactions of dinitrobenzoannulated heterocycles (furazan, thiadiazole, selenadiazole, pyridine) with anionic C-nucleophiles (mono- and diketones, nitroalkanes and related compounds) provided stable anionic adducts in high yields. Consecutive oxidation with ammonium cerium (IV) nitrate resulted in re-aromatization with the formation of the corresponding substitution products, formally representing C-H-functionalized benzoheterocycles.

**Keywords:** CH-functionalization; nitroarenes; carbon nucleophiles; nucleophilic addition; anionic  $\sigma$ -complexes

## 1. Introduction

The functionalization of arenes provides their chemical diversity and opens a way to valuable substances that are widely used in medicine, pharmaceuticals, agriculture and other areas. In recent years, the functionalization of the CH bond has become an important tool for the implementation of such processes. Nucleophilic substitution of hydrogen ( $S_N^H$ ) in arenes has acquired intensive development as a more prospective way of functionalization than classical  $S_NAr$  processes occurring through ipso-substitution of a nucleofuge. Two general  $S_N^H$  processes are oxidative (ONSH) and vicarious (VNSH) nucleophilic substitution of hydrogen proceeding through the generation of  $\sigma^H$ -adduct. In the case of highly electrophilic substrates, the intermediate sigma-adducts can be isolated and identified, and their chemical behavior can be studied.

It is well-known that highly electrophilic arenes and heteroarenes (super-electrophiles) readily form adducts with nucleophiles of various nature, including weak neutral nucleophiles such as  $\pi$ -excessive (het)arenes, enamines, etc. [1–3]. In the case of C-nucleophiles, these adducts can be isolated. Earlier we reported on the reactions of some azolo[b]pyridines with 1,3-dicarbonyl compounds [4–6]. In this work, we studied the reactions of dinitrobenzoannulated heterocycles (thiadiazole, selenadiazole, pyridine) with anionic C-nucleophiles (mono- and diketones, nitroalkanes and related compounds).

## 2. Results and Discussion

Among numerous highly electrophilic nitro (het)arenes, the following were selected for this study: 4,6-dinitrobenzothiadiazole **1a** [7], 4,6-dinitrobenzoselenadiazole **1b** [8], 5,7-dinitroquinoline **1c** [9] and 5,7-dinitroquinoline-*N*-oxide **1d** [9]. It was found that their reactions with mono- and diketones as well as 2-nitropropane in the presence of a base provided the previously unknown stable anionic adducts **2** in high yields (Figure 1, Table 1). These adducts were isolated in pure form and are characterized by (nuclear magnetic resonance spectroscopy (NMR) and high-resolution mass spectrometry

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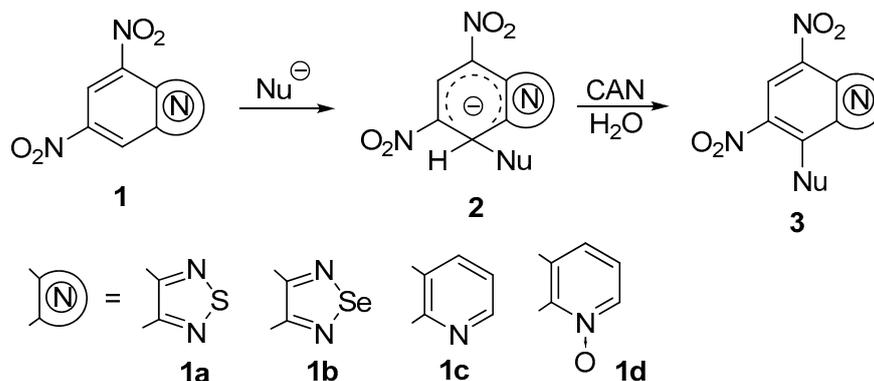
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(HRMS). Oxidation of compounds **2** with ammonium cerium (IV) nitrate was studied. In some cases (Table 1, entries 1, 2, 6, 8) the corresponding substitution products **3** were isolated formally representing C-H-functionalized benzoheterocycles. In case of adduct **2c**, decomposition was observed, while in case of dinitroquinoline complex **2i**, the starting compound **1c** appeared to be the sole isolable product. In all other cases,  $^1\text{NMR}$  spectra showed a 1:1 mixture of the target substitution product and the corresponding starting material (Table 1, entries 4, 5, 7, 10, 11).



Reaction conditions:

nitro compound **1** (1 mmol), ketone (5 mL),  $\text{Et}_3\text{N}$  (1 mmol),  $20^\circ\text{C}$ , 1h  
 or nitro compound **1** (1 mmol), 2-nitropropane (5 mL),  $t\text{-BuOK}$  (1 mmol),  $20^\circ\text{C}$ , 1h  
 Oxidation: complex **2** (1 mmol),  $(\text{NH}_4)_2\text{Ce}(\text{NO}_3)_6$  (2 mmol),  $\text{H}_2\text{O}$  (10 mL),  $20^\circ\text{C}$ , 10 min.

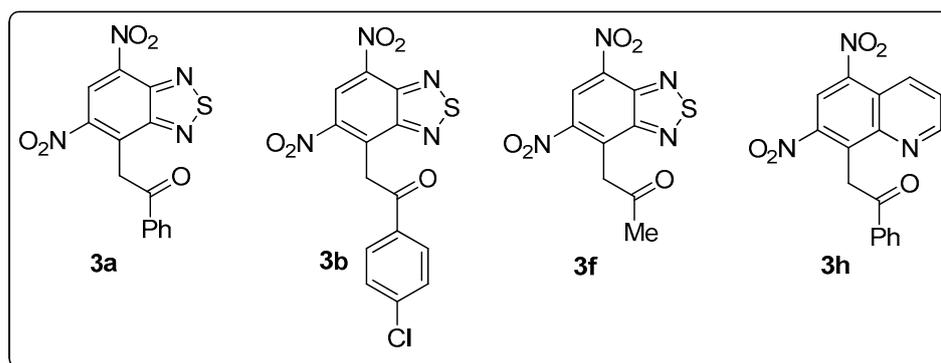
Figure 1. Formation of anionic complexes with C-nucleophiles and their oxidation.

Table 1. Yields of anionic  $\sigma$ -adducts **2** and oxidation products **3**.

Entry	Compound 1	NuH	Compound 2, Yield (%)	Compound 3, Yield (%)
1	<b>1a</b>	PhCOMe	<b>2a</b> , 88	<b>3a</b> , 47
2	<b>1a</b>	4-Cl-C <sub>6</sub> H <sub>4</sub> COMe	<b>2b</b> , 84	<b>3b</b> , 36
3	<b>1a</b>	Me <sub>2</sub> CHNO <sub>2</sub>	<b>2c</b> , 64	- <sup>2</sup>
4	<b>1a</b>	Cyclohexanone	<b>2d</b> , 67	- <sup>3</sup>
5	<b>1a</b>	MeCOCH <sub>2</sub> COMe	<b>2e</b> , 87	- <sup>3</sup>
6	<b>1a</b>	Acetone	<b>2f</b> , 82	<b>3f</b> , 46
7	<b>1b</b>	Me <sub>2</sub> CHNO <sub>2</sub>	<b>2g</b> , 92	- <sup>3</sup>
8	<b>1c</b>	PhCOMe	<b>2h</b> , - <sup>1</sup>	<b>3h</b> , 28
9	<b>1c</b>	Me <sub>2</sub> CHNO <sub>2</sub>	<b>2i</b> , 98	- <sup>4</sup>
10	<b>1d</b>	PhCOMe	<b>2j</b> , 77	- <sup>3</sup>
11	<b>1d</b>	Me <sub>2</sub> CHNO <sub>2</sub>	<b>2k</b> , 24	- <sup>3</sup>

<sup>1</sup> Compound **2h** was not isolated in pure form and was oxidized without further purification. <sup>2</sup> Decomposition. No identified product was isolated. <sup>3</sup> A 1:1 mixture of oxidation product **3** and starting compound **1** was formed. <sup>4</sup> Compound **1c** was recovered (20%) along with unidentified decomposition products.

As it follows from the data presented in Table 1, the oxidation of adducts **2** generally proceed in two directions: formation of target substitution products **3** and decomposition to give starting compounds **1**. Such behavior of anionic  $\sigma$ -complexes is not surprising since it was reported earlier for 1,3,5-trinitrobenzene (TNB) adducts with CH-acidic compounds [10]. The kinetic study of the decomposition of the TNB-acetophenone complex revealed a strong dependence on the pH of the reaction media. However, in a number of cases we were able to isolate polyfunctional derivatives of highly electrophilic benzoannulated heterocycles (Figure 2).



**Figure 2.** Compounds synthesized by oxidation of anionic  $\sigma$ -complexes of dinitrobenzohetarenes and C-nucleophiles.

### 3. Experimental Procedures

**Anionic  $\sigma$ -adducts 2a,b,d-f,h,j (general procedure).** To a solution of dinitro compound **1** in an appropriate ketone (5 mL), Et<sub>3</sub>N (0.14 mL, 1 mmol) was added. The mixture was stirred for 1 h at 20 °C, poured in ether (50 mL), and the resulting precipitate was filtered off, washed with ether and dried to give the target adduct (see Table 1 for yields).

**Anionic  $\sigma$ -adducts 2c,g,i,k (general procedure).** To a solution of dinitro compound **1** in 2-nitropropane (5 mL), t-BuOK (0.112 g, 1 mmol) was added. The mixture was stirred for 1 h at 20 °C, poured in ether (50 mL), and the resulting precipitate was filtered off, washed with ether and dried to give target adduct (see Table 1 for yields).

**Oxidation of adducts 2 (general procedure).** To a solution of the corresponding adduct **2** (1 mmol) in 5 mL of H<sub>2</sub>O, a solution of (NH<sub>4</sub>)<sub>2</sub>Ce(NO<sub>3</sub>)<sub>4</sub> (1.1 g, 2 mmol) in H<sub>2</sub>O (5 mL) was added. The mixture was stirred for 10 min at 20 °C and extracted with CHCl<sub>3</sub> (3 × 10 mL). Organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated to give target compound **3** which was then purified by column chromatography (SiO<sub>2</sub>, CHCl<sub>3</sub>) (see Table 1 for yields).

### 4. Conclusions

Thus, a series of the previously unknown dinitrobenzofuroxans and azines functionalized in a benzene ring were synthesized using stable anionic  $\sigma$ -adducts as key intermediates of the C-H functionalization of  $\pi$ -deficient nitroarenes.

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