

## Supplementary Materials:

# Synthesis, Structure and Sodium Mobility of Sodium Vanadium Nitridophosphate: A Zero-Strain and Safe High Voltage Cathode Material for Sodium-ion Batteries

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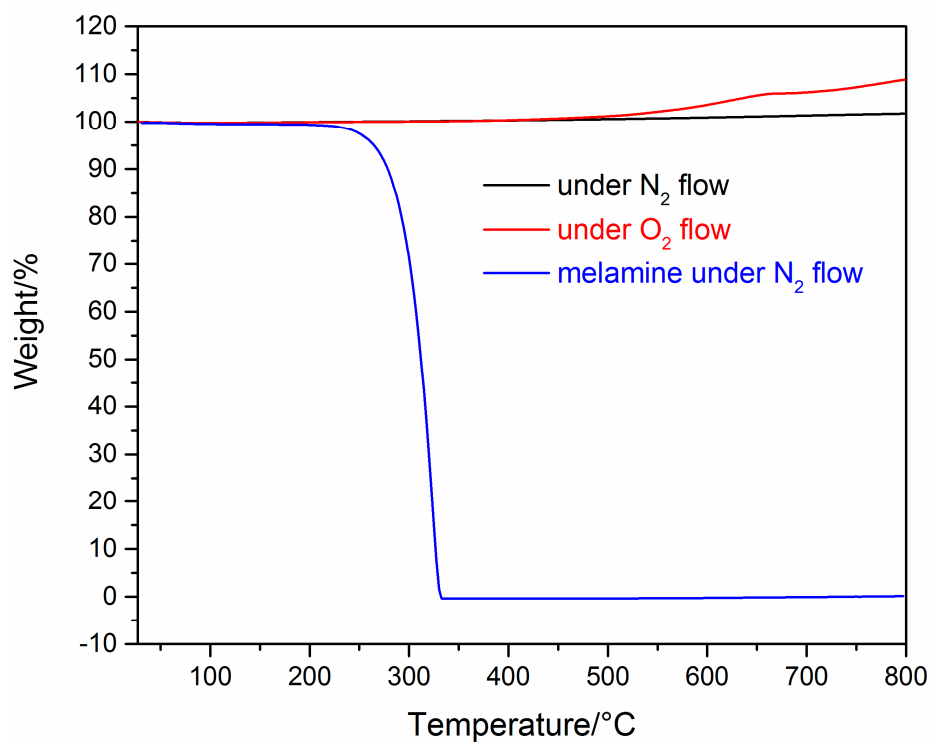
**Table S1.** Crystallographic data of  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$ .

	Value
Chemical formula	$\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$
Crystal system, space group	Cubic, $P2_13$
R-bragg	9.7%
Cell mass	1483.3
Cell volume ( $\text{\AA}^3$ )	841.3 (9)
Crystal density ( $\text{g cm}^{-3}$ )	2.9307 (3)
Temperature	~298 K (ambient)
Lattice parameters	-
$a$ ( $\text{\AA}$ )	9.440 (1)
$R_{\text{wp}}$	8.7%
$R_{\text{p}}$	6.7%
GOF	1.41
Radiation type	Cu $K_{\alpha}$ , $\lambda = 1.541(8)$ $\text{\AA}$

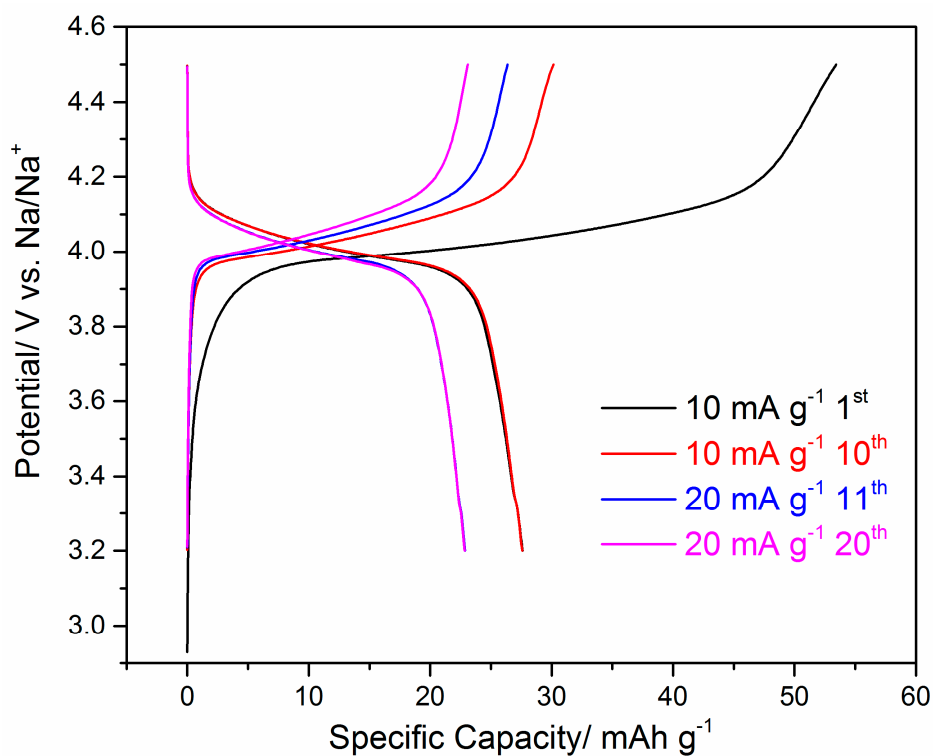
**Table S2.** Atomic parameters for  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$ .

Site	$x^*$	$y^*$	$z^*$		Occ*	$B_{\text{eq}}^*$
Na1	-0.26140	0.73860	0.73860	$\text{Na}^+$	1	0.0255
Na2	0.14230	0.64230	0.85770	$\text{Na}^+$	1	0.0181
Na3	-0.05180	0.94820	0.94820	$\text{Na}^+$	1	0.029
V1	0.33070	0.83070	0.66930	$\text{V}^{3+}$	1	0.0109
P1	0.00415	0.83568	0.58270	$\text{P}^{5+}$	1	0.0092
N1	0.05150	0.94850	0.44850	$\text{N}^{3-}$	1	0.0087
O1	-0.09860	0.72930	0.52000	$\text{O}^{2-}$	1	0.0154
O2	0.13720	0.75680	0.62370	$\text{O}^{2-}$	1	0.0124
O3	-0.06150	0.91810	0.70020	$\text{O}^{2-}$	1	0.017

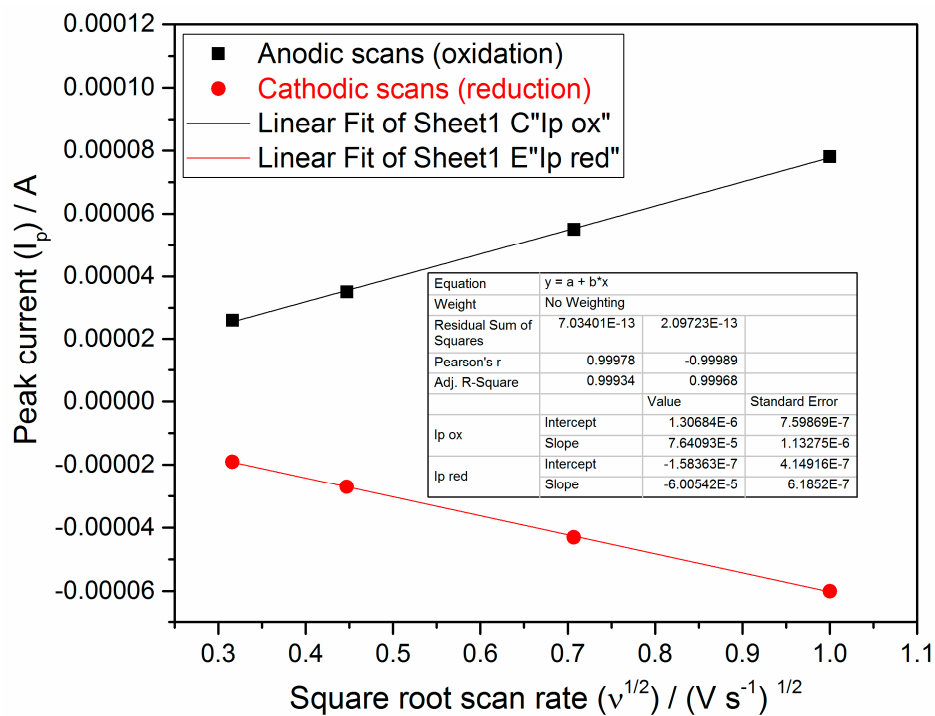
\* fixed parameters



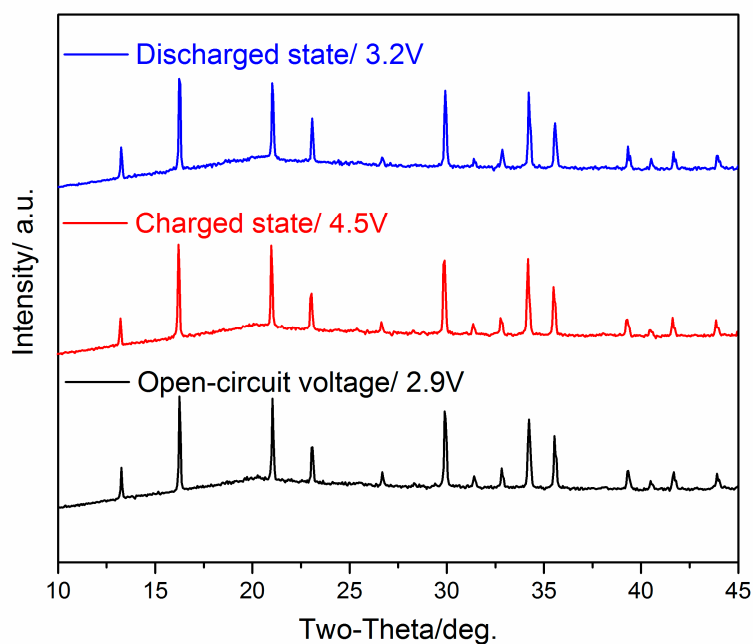
**Figure S1.** TG curves of  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  under  $\text{N}_2$  and  $\text{O}_2$  flows and melamine under  $\text{N}_2$  flow of  $20 \text{ ml min}^{-1}$  within the temperature range of  $30\text{--}800 \text{ }^\circ\text{C}$  at  $5 \text{ }^\circ\text{C min}^{-1}$  heating rate.



**Figure S2.** Galvanostatic charge-discharge profiles of  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  at current density of  $10 \text{ mA g}^{-1}$  for the 1st and 10th cycles and  $20 \text{ mA g}^{-1}$  for 11th and 20th cycles.



**Figure S3.** Peak current ( $I_p$ , A  $cm^{-2}$ ) vs. square root of scan rate ( $v^{1/2}$ ,  $V^{1/2} s^{-1/2}$ ) and related linear fit corresponding to the  $V^{3+/4+}$  redox processes.



**Figure S4.** In-situ X-ray diffraction patterns of  $Na_3V(PO_3)_3N$  at OCV, after the 1st charge (4.5 V) and 1st discharge state (3.2 V).

