

Short Note

N-(2-(1*H*-indol-3-yl)ethyl)-2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanamide

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Abstract: *N*-(2-(1*H*-Indol-3-yl)ethyl)-2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanamide was prepared by a reaction between tryptamine and flurbiprofen, applying *N,N'*-dicyclohexylcarbodiimide, as a coupling agent. The obtained new amide has a fragment similar to *Brequinar*, a compound used in SARS-CoV-2 treatment trials. The newly synthesized compound was fully analyzed and characterized via ¹H, ¹³C-NMR, UV, IR, and mass spectral data.

Keywords: amide; flurbiprofen; SARS-CoV-2; tryptamine

1. Introduction

One of the most potent currently known 2-arylpropionic acids is flurbiprofen, which has anti-inflammatory, analgesic, and antipyretic effects [1]. Flurbiprofen is widely used as an anti-inflammatory drug, both orally and topically, for the symptomatic treatment of chronic inflammatory diseases such as rheumatoid arthritis and osteoarthritis. Flurbiprofen is generally used as a racemate of (*R*)- and (*S*)-enantiomers. For the inhibition of cyclooxygenase activity, (*S*)-flurbiprofen is responsible. Along with the positives, the use of flurbiprofen can cause many side effects, such as abdominal cramps and pain, diarrhea, dyspepsia, edema, headache, and nausea [2]. Biphenyl substituent is very well known for owing biological activity. Flurbiprofen has the same biphenyl nucleus as *Brequinar*, differing only in the position of the fluorine atom (Figure 1). From the literature, it is known that *N*-contained six-membered cycles with biphenyl substituent are inhibitors of dihydroorotate dehydrogenase (DHODH) [3]. The inhibitors of DHODH have immunosuppressant, antiproliferative, and antimalarial activities, e.g., DSM265 [4]. *Brequinar* is a well known inhibitor of human DHODH, which is used in some pathologies in clinical trials including newly emerged coronavirus SARS-CoV-2 [5–7].

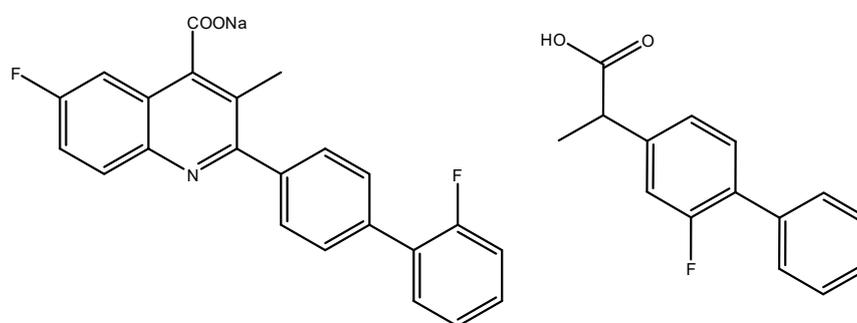


Figure 1. Structural formulas of *Brequinar* (left) and *Flurbiprofen* (right).

The fluorine substitution of biologically active molecules exercises an influence on their properties and activities. Fluorine substitution in a drug molecule can influence not



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only pharmacokinetic properties, such as absorption, tissue distribution, excretion, and the route and rate of biotransformation, but also its pharmacodynamics and toxicology [8].

Tryptamine is a biogenic amine, naturally occurring in plants, animals, and microorganisms [9]. It also belongs to the so-called “trace amine” group, compounds present in all vertebrate and invertebrate species, typically in the central nervous system [10]. Due to the diverse pharmacological properties of tryptamine and the proven anti-inflammatory properties of flurbiprofen, it is of great interest to create a molecule that combines the two molecules together and improves their properties. Rose and co-authors reported the synthesis of serotonin derivatives containing NSAIDs (Nonsteroidal anti-inflammatory drugs) in their structures [11]. In Figure 2 is presented the structural formula of the serotonin derivative of flurbiprofen.

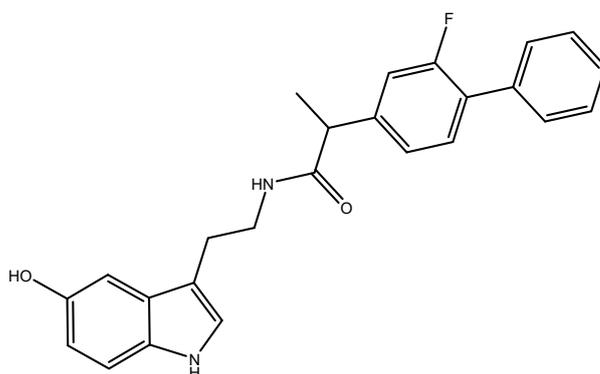
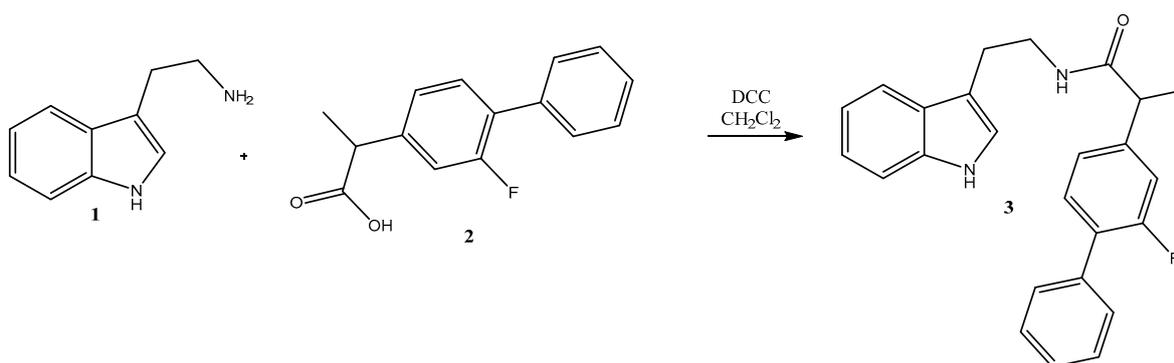


Figure 2. Structural formula of 2-(2-fluoro-[1,1'-biphenyl]-4-yl)-N-(2-(5-hydroxy-1H-indol-3-yl)ethyl)propanamide.

Due to the importance of the amides in pharmaceutical synthesis [12,13], a coupling between flurbiprofen and tryptamine via amide bond formation was achieved in order to obtain *N*-(2-(1H-indol-3-yl)ethyl)-2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanamide **3**.

2. Results

Herein, we report the successfully synthesized *N*-(2-(1H-indol-3-yl)ethyl)-2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanamide **3**, as shown in Scheme 1.



Scheme 1. Synthesis of *N*-(2-(1H-indol-3-yl)ethyl)-2-(2-fluoro-[1,1'-biphenyl]-4-yl)propanamide **3**.

An easy synthetic procedure for amide synthesis is the *N,N'*-dicyclohexylcarbodiimide (DCC)-mediated coupling between carboxylic acids and amines. DCC is commonly used for the preparation of esters, amides, or anhydrides. DCC reacts with the carboxyl group of flurbiprofen to produce an activated acylating agent that reacts with the amino group of the other molecule to form an amide bond.

The resultant compound is characterized by its melting point, ^1H and ^{13}C -NMR, UV, IR, and HRMS spectra.

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Data Availability Statement: The data presented in this study are available in this article and supporting supplementary material.

Conflicts of Interest: The authors declare no conflict of interest.

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