

Short Note

# (2*S*\*,4*S*\*)-4-[(*E*)-(2,2-Dimethylhydrazono)methyl]-6-methoxy-4-methyl-2-[(*E*)-styryl]-1,2,3,4-tetrahydroquinoline

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**Abstract:** The Povarov reaction of *p*-anisidine, cinnamaldehyde and methacrolein dimethylhydrazono afforded a 1,2,3,4-tetrahydroquinoline derivative bearing 2-styryl, 4-methyl and 4-dimethylhydrazono substituents in a fully diastereoselective fashion. This is the first example of the combination of a type I aza-vinylogous Povarov reaction and a type II vinylogous Povarov reaction in the same process.

**Keywords:** nitrogen heterocycles; tetrahydroquinolines; styrylquinolines; Povarov reaction



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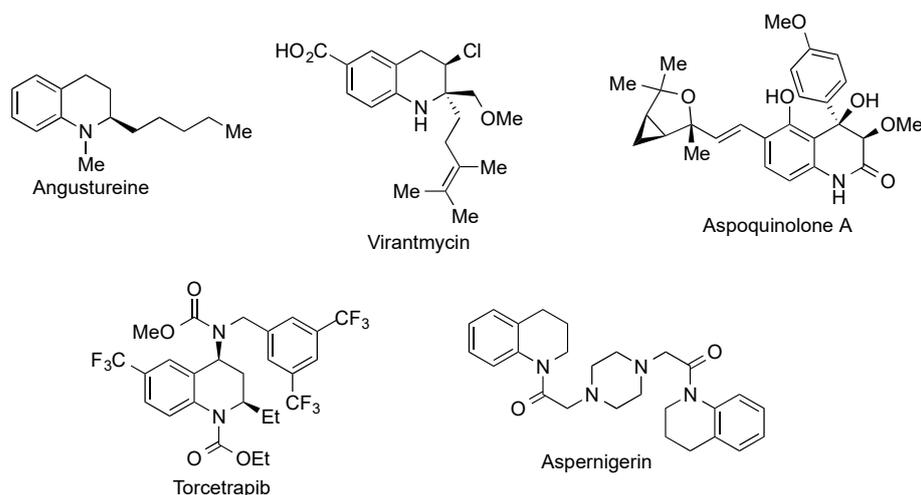
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## 1. Introduction

The 1,2,3,4-tetrahydroquinoline ring system is present in a broad variety of natural products and synthetic bioactive molecules [1,2] and can be considered as one of the most relevant simple heterocycles. Some important tetrahydroquinolines are summarized in Figure 1, including the alkaloids angustureine, virantmycin and aspoquinolone A, the cholesterol-lowering agent torcetrapib and the insecticidal, herbicidal and fungicidal aspernigerin (Figure 1).

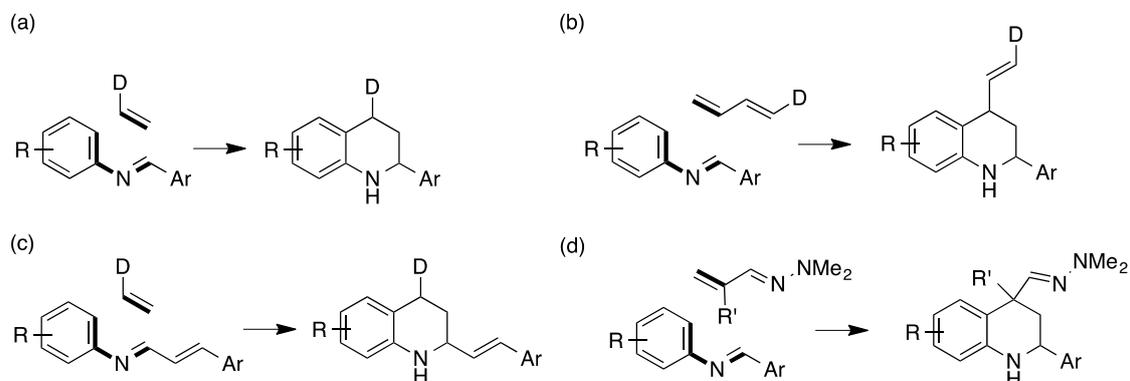


**Figure 1.** Structures of selected tetrahydroquinoline derivatives.

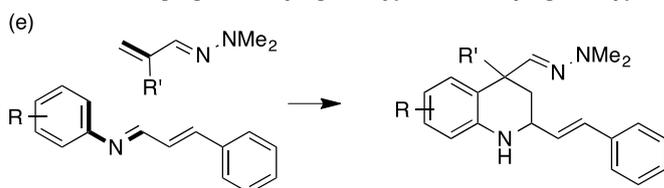
The importance of tetrahydroquinolines has prompted a great deal of research into their synthesis. Among the many known approaches to this framework [1–3], the Povarov reaction, i.e., the acid-catalyzed formal inverse electron demand [4+2] cycloaddition of electron-rich olefins with *N*-arylimines, arising from arylamines and aldehydes (Scheme 1a), stands out as one of the most studied methods [4,5]. Vinylogous Povarov reactions are also known, and have the advantage of generating tetrahydroquinolines with an olefin substituent either at C-4 (Type I, Scheme 1b) or C-2 (Type II, Scheme 1c). We have

described [6,7] a type I aza-vinylogous Povarov reaction using  $\alpha,\beta$ -unsaturated hydrazones as the dienophile component that has the advantage of simultaneously installing a quaternary stereocenter and functional group at C-4 (Scheme 1d).

**Literature precedent: Standard and vinylogous Povarov reactions**



**This work: Merging aza-vinylogous Type I and vinylogous Type II Povarov reactions**

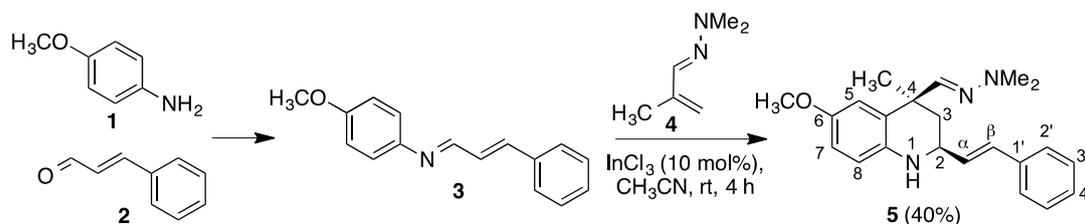


**Scheme 1.** (a) The standard Povarov reaction; (b) type I vinylogous Povarov reaction; (c) type II vinylogous Povarov reaction; (d) type I aza-vinylogous Povarov reaction; (e) the reaction described in this work, which combines the features of the type I aza-vinylogous and type II vinylogous Povarov reactions.

In this context, we describe here the first example of a Povarov reaction that combines the features of a type I aza-vinylogous and a type II vinylogous Povarov reaction in a single transformation and allows the preparation of 2,4-difunctionalized 2-styryltetrahydroquinolines bearing a quaternary stereocenter at C-4 (Scheme 1e).

## 2. Results and Discussion

The doubly vinylogous Povarov reaction that we describe here is summarized in Scheme 2. The reaction between *p*-anisidine **1** and cinnamaldehyde **2** afforded the corresponding imine **3**, which was treated in crude state with methacrolein dimethylhydrazone **4** in acetonitrile containing 10% indium trichloride as a Lewis acid catalyst, at room temperature, affording compound **5** as a single diastereomer in 40% overall yield (Scheme 2).



**Scheme 2.** Synthesis of compound **5**.

The structure of compound **5** is consistent with a high-resolution mass measurement and with its IR,  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectral data, which were assigned with the aid of 2D-NMR experiments. Thus, the hydrazone group gave a  $\text{CH}=\text{N}$  stretching vibration at  $1600\text{ cm}^{-1}$  in the IR spectrum, a  $^1\text{H-NMR}$  signal in the 6.74–6.64 interval, which overlapped

with other signals but is clearly visible in the HMQC experiment, and a  $^{13}\text{C}$ -NMR signal at 145.3 ppm. One of the olefinic protons was clearly observed at 6.27 ppm as a doublet of doublets with  $J = 15.8$  and  $7.3$  Hz, which allowed its assignment as H- $\alpha$ . The other proton corresponding to the olefin part of the styryl group (H- $\beta$ ) is part of the multiplet at 6.74–6.64, as revealed by the COSY experiment, and the H-C correlation experiments allowed the assignment of the olefin carbons to the CH signals at 132.4 (C- $\alpha$ ) and 131.0 (C- $\beta$ ). The *cis* arrangement of the styryl and dimethylhydrazono substituents agrees with the literature precedents [6,7] and was unequivocally established by the observation of a NOE enhancement of the axial C-4 methyl substituent upon irradiation of the H-2 proton (Figure 2). The alternative *trans* isomer was not observed in the crude reaction product.

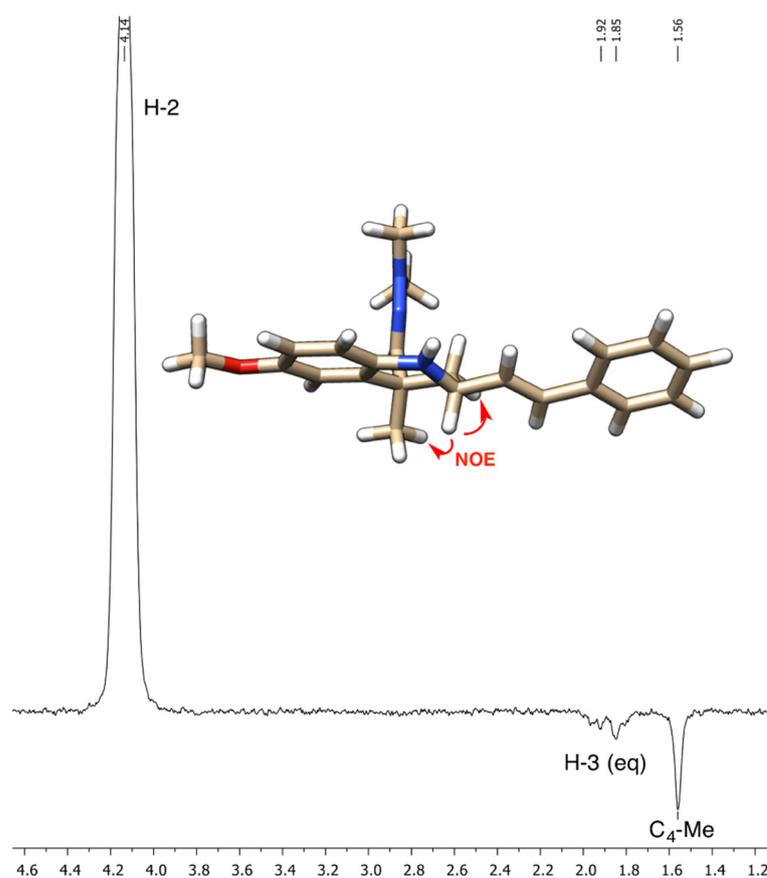


Figure 2. NOE assignment of the relative configuration of compound 5.

### 3. Materials and Methods

**General experimental information.** All reagents (Sigma-Aldrich, Madrid, Spain; Fischer Chemical, Madrid, Spain; Alpha Aesar, Kandel, Germany) and solvents (Scharlau, Barcelona, Spain; Fischer Chemical, Madrid, Spain) were of commercial quality and were used as received. Reactions were monitored by thin layer chromatography on aluminum plates coated with silica gel and fluorescent indicator (Merck, Madrid, Spain). Infrared spectra were recorded with an Agilent Cary630 FTIR spectrophotometer (Madrid, Spain) working by attenuated total reflection (ATR), with a diamond accessory for solid and liquid samples. NMR spectroscopic data were recorded using a Bruker Avance 250 spectrometer (Bruker, Rivas-Vaciamadrid, Spain) operating at 250 MHz for  $^1\text{H}$ -NMR and 63 MHz for  $^{13}\text{C}$ -NMR maintained by the NMR facility of Universidad Complutense (CAI de Resonancia Magnética Nuclear, Madrid, Spain); chemical shifts are given in ppm and coupling constants in Hertz.  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR assignments were supported by 2D-NMR experiments and were aided by simulations performed with MestreNova and ChemDraw Pro. Copies of spectra and 2D-NMR experiments are provided in the Supporting Information. Time-

of-flight mass spectrometric measurements were performed using a MALDI-TOF/TOF Bruker ULTRAFLEX (mass range: 300–150,000 u) at the CAI of Espectrometría de Masas, Universidad Complutense.

**(E)-4-methoxy-N-((E)-3-phenylallylidene)aniline (3).** A solution of *p*-anisidine (1 mmol, 123 mg) and cinnamaldehyde (1 eq, 148 mg) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) was stirred vigorously in the presence of anhydrous Na<sub>2</sub>SO<sub>4</sub> (5 g) for 30 min and then the reaction mixture was filtered and the solvent evaporated under vacuum to afford **3** as a brown foam in quantitative yield. The crude was used in the following reaction without further purification. IR (neat): 2955.4 (C-H), 1626.0 (C=N), 1601.3 (C=C), 1498.8 (Csp<sup>2</sup>-Csp<sup>2</sup>), 1242.3 (Csp<sup>2</sup>-O) cm<sup>-1</sup>. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ: 8.31 (dd, *J* = 5.9, 2.4 Hz, 1H), 7.62–7.52 (m, 2H), 7.47–7.32 (m, 3H), 7.32–7.21 (m, 2H), 7.20–7.11 (m, 2H), 7.01–6.91 (m, 2H), 3.84 (s, 3H) ppm. <sup>13</sup>C{<sup>1</sup>H} NMR (63 MHz, CDCl<sub>3</sub>) δ: 159.8, 158.7, 144.8, 143.4, 136.1, 129.7, 129.2, 129.1, 127.7, 122.6, 114.7, 55.8.

**(±)-(2S\*,4S\*)-4-((E)-(2,2-dimethylhydrazono)methyl)-6-methoxy-4-methyl-2-((E)-styryl)-1,2,3,4-tetrahydroquinoline (5).** To a stirred solution of compound **3** (1.0 mmol, 237 mg) and InCl<sub>3</sub> (0.1 mmol, 22 mg) in acetonitrile (20 mL) was added dropwise methacrolein dimethylhydrazone (1.1 mmol, 123 mg). Stirring at room temperature was continued until completion of the reaction, as indicated by TLC (R<sub>f</sub> 0.29, eluting with 8:2 hexane/ethyl acetate), and the mixture was then diluted with water (10 mL), extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 × 10 mL), dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure. The resulting crude was purified by silica gel column chromatography using hexane/ethyl acetate (92.5:7.5) as the mobile phase to obtain 140 mg (40%) of compound **5** as a yellow oil. IR (neat): 3369.8 (N-H), 2951.1 (C-H), 2826.0 (C-H), 1600.2 (C=N), 1501.6 (C=C), 1235.3 (Csp<sup>2</sup>-O) cm<sup>-1</sup>. <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ: 7.45–7.27 (m, 5H, Ph), 6.74–6.64 (m, 4H, H-5, H-7, CH=N, H-β), 6.57 (d, *J* = 9.1 Hz, 1H, H-8), 6.27 (dd, *J* = 15.8, 7.3 Hz, 1H, H-α), 4.19–4.08 (m, 1H, H-2), 3.76 (s, 3H, OCH<sub>3</sub>), 2.79 (s, 6H, N(CH<sub>3</sub>)<sub>2</sub>), 2.05–1.90 (m, 1H, H-3<sub>ax</sub>), 1.82 (dd, *J* = 13.1, 2.9 Hz, 1H, H-3<sub>eq</sub>), 1.56 (s, 3H, C<sub>4</sub>-CH<sub>3</sub>) ppm; NH is absent. <sup>13</sup>C{<sup>1</sup>H} NMR (63 MHz, CDCl<sub>3</sub>) δ: 152.6 (C<sub>q</sub>, C-6), 145.3 (CH=N), 138.2 (C<sub>q</sub>, C-8a), 137.2 (C<sub>q</sub>, C-1'), 132.4 (C-α), 131.0 (C-β), 129.1 (C-3', C-5'), 128.7 (C<sub>q</sub>, C-4a), 128.1 (C-4'), 126.8 (C-2', C-6'), 116.1 (C-8), 114.7 (C-7), 113.9 (C-5), 56.3 (OCH<sub>3</sub>), 51.6 (C-2), 43.9 (N(CH<sub>3</sub>)<sub>2</sub>), 41.9 (C-3), 41.0 (C<sub>q</sub>, C-4), 28.5 (C<sub>4</sub>-CH<sub>3</sub>) ppm. HRMS (MALDI-TOF): calculated for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O, 349.2154. Found, 349.2161.

#### 4. Conclusions

We describe the first example of a doubly vinylogous Povarov reaction involving an α,β-unsaturated aromatic imine as diene and an α,β-unsaturated dimethylhydrazone as the dienophile. This reaction proceeded with complete diastereoselectivity and afforded a 1,2,3,4-tetrahydroquinoline derivative bearing a C-4 quaternary stereocenter and *cis*-arranged functional groups at C-4 (dimethylhydrazono) and C-2 (styryl).

**Supplementary Materials:** The following are available online. Copies of spectra of compounds **3** and **5**.

**Author Contributions:** Conceptualization, J.C., M.T.R. and J.C.M.; methodology, J.C.; writing—original draft preparation, J.C.M.; writing—review and editing, J.C., M.T.R. and J.C.M.; supervision, M.T.R. and J.C.M.; funding acquisition, J.C.M. All authors have read and agreed to the published version of the manuscript.

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## References

1. Sridharan, V.; Suryavanshi, P.; Menéndez, J.C. Advances in the chemistry of tetrahydroquinolines. *Chem. Rev.* **2011**, *111*, 7157–7259. [[CrossRef](#)] [[PubMed](#)]
2. Muthukrishnan, I.; Sridharan, V.; Menéndez, J.C. Progress in the chemistry of tetrahydroquinolines. *Chem. Rev.* **2019**, *119*, 5057–5191. [[CrossRef](#)] [[PubMed](#)]
3. Katritzky, A.R.; Rachwal, S.; Rachwal, B. Recent progress in the synthesis of 1,2,3,4-tetrahydroquinolines. *Tetrahedron* **1996**, *52*, 15031–15070. [[CrossRef](#)]
4. Kouznetsov, V.V. Recent synthetic developments in a powerful imino Diels–Alder reaction (Povarov reaction): Application to the synthesis of N-polyheterocycles and related alkaloids. *Tetrahedron* **2009**, *65*, 2721–2750. [[CrossRef](#)]
5. Ghashghaei, O.; Masdeu, C.; Alonso, C.; Palacios, F.; Lavilla, R. Recent advances of the Povarov reaction in medicinal chemistry. *Drug Discov. Today Technol.* **2018**, *29*, 71–79. [[CrossRef](#)] [[PubMed](#)]
6. Sridharan, V.; Perumal, P.T.; Avendaño, C.; Menéndez, J.C. The first aza Diels–Alder reaction involving an  $\alpha,\beta$ -unsaturated hydrazone as the dienophile: Stereoselective synthesis of C-4 functionalized 1,2,3,4-tetrahydroquinolines containing a quaternary stereocenter. *Org. Biomol. Chem.* **2007**, *5*, 1351–1353. [[CrossRef](#)] [[PubMed](#)]
7. Sridharan, V.; Ribelles, P.; Estévez, V.; Villacampa, M.; Ramos, M.T.; Perumal, P.T.; Menéndez, J.C. New types of reactivity of  $\alpha,\beta$ -unsaturated N,N-dimethylhydrazones: Chemodivergent, diastereoselective synthesis of functionalized tetrahydroquinolines and hexahydropyrrolo[3,2-b]indoles. *Chem. Eur. J.* **2012**, *18*, 5056–5063. [[CrossRef](#)]