

*Supplemental files*

## **Interaction of ethylene with $\text{Ir}_n$ (n = 1-10): From bare clusters to $\gamma\text{-Al}_2\text{O}_3$ -supported nanoparticles**

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**Table S1:** Geometry, magnetic moment ( $M$ ), and energy of gas phase  $\text{Ir}_n$  ( $n = 2-10$ ) clusters.

n	Geometry	$M$ [ $\mu_B$ ]	$d(\text{Ir-Ir})$ [ $\text{\AA}$ ]	CE <sup>a</sup> [eV/atom]	BE <sup>b</sup> [eV]
2	$D_{\infty h}$	4(4 <sup>c</sup> )	2.21(2.21 <sup>c</sup> )	-2.68(-2.61 <sup>c</sup> )	-5.36
3	$D_{\infty h}$	1(1 <sup>c</sup> )	2.18(2.18 <sup>c</sup> )	-3.36(-3.27 <sup>c</sup> )	-5.03
4	$D_{4h}$	4(4 <sup>c,d</sup> )	2.31(2.31 <sup>c</sup> )	-3.92(-3.85 <sup>c</sup> , 3.77 <sup>d</sup> )	-3.92
5	$C_{4v}$	5(5 <sup>c,e</sup> )	2.47(2.47 <sup>c</sup> )	-4.16(-4.08 <sup>c</sup> , 4.01 <sup>d</sup> )	-2.60
6	$D_{3h}$	6(6 <sup>c,e</sup> )	2.42(2.42 <sup>c</sup> )	-4.54(-4.45 <sup>c</sup> , 4.38 <sup>d</sup> )	-3.03
7	$C_{2v}$	11(11 <sup>c,d,e</sup> )	2.51(2.51 <sup>c</sup> )	-4.64(-4.56 <sup>c</sup> , 4.51 <sup>d</sup> )	-2.50
8	$O_h$	0(0 <sup>c,d,e</sup> )	2.37(2.37 <sup>c</sup> )	-5.08(-5.00 <sup>c</sup> , 4.90 <sup>d</sup> )	-3.39
9	$C_s$	3(3 <sup>c,e</sup> )	2.38(2.39 <sup>c</sup> )	-5.08(-5.00 <sup>c</sup> , 4.90 <sup>d</sup> )	-3.27
10	$C_{2v}$	4(4 <sup>d,e</sup> )	2.42(2.42 <sup>c</sup> )	-5.23(-5.15 <sup>c</sup> , 5.02 <sup>d</sup> )	-3.08

<sup>a</sup> Cohesive energy =  $(E[\text{Ir}_n] - nE[\text{Ir}])/n$

<sup>b</sup> Bond energy =  $n\text{CE}/m$  where  $m$  is the number of Ir-Ir bonds in the cluster

<sup>c</sup> Ref.[1], VASP code, PW91 functional, 380 eV;

<sup>d</sup> Ref.[2], VASP code, PW91 functional, 300 eV;

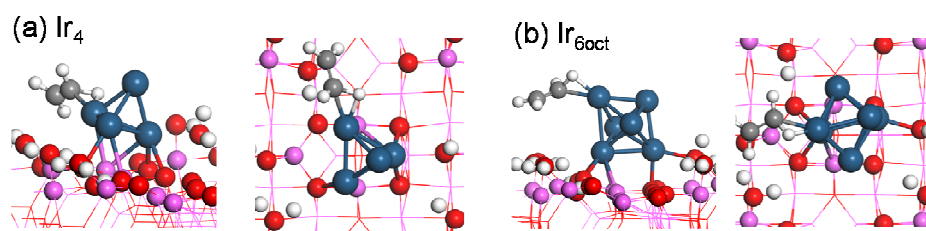
<sup>e</sup> Ref.[3], Dmol<sup>3</sup> code, BPW91 functional

**Table S2:** Adsorption energy  $E_{\text{ads}}$  (eV) and Nucleation energy  $E_{\text{nuc}}$  (eV) for  $\text{Ir}_n$  clusters on  $\gamma\text{-Al}_2\text{O}_3$  surfaces.

N	$E_{\text{ads}}$		$E_{\text{nuc}}$	
	$\gamma\text{-Al}_2\text{O}_3(110)$	$\gamma\text{-Al}_2\text{O}_3(001)^{\text{a}}$	$\gamma\text{-Al}_2\text{O}_3(110)$	$\gamma\text{-Al}_2\text{O}_3(001)^{\text{a}}$
1	-2.58	--	--	--
2	-2.53	-1.03	-2.73	-2.77
3	-3.67	-1.17	-3.01	-2.75
4	-4.12	-2.65	-3.00	-4.31
5	-3.74	-2.18	-2.92	-4.04
6	-3.92	-0.81	-4.05	-3.19
7	-3.38	-1.82	-2.18	-4.44
8	-2.16	-1.96	-4.33	-6.48
9	-2.74	-1.64	-3.06	-3.33
10	-2.98	-1.78	-4.25	-4.71

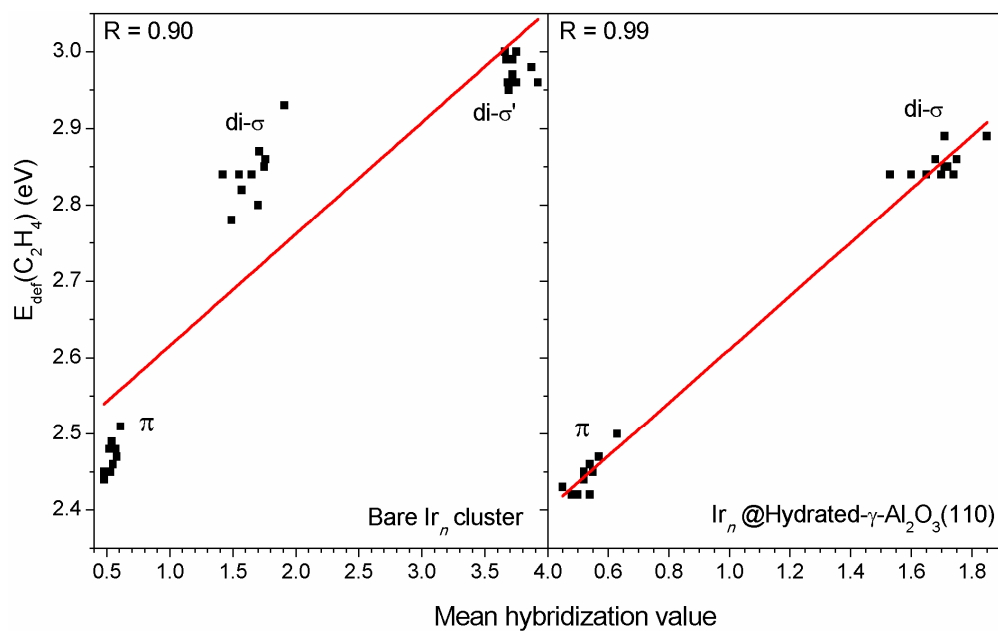
<sup>a</sup> Ref.[1], VASP code, PW91 functional, cutoff energy 380 eV.

## M configuration of ethylene adsorption



**Figure S1:** Most stable M configuration of ethylene adsorption on hydrated  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>(110)-supported (a)Ir<sub>4</sub> and Ir<sub>6oct</sub>.

Linear fitting of ethylene deformation energy  $E_{\text{def}}(\text{C}_2\text{H}_4)$  (eV) and mean hybridization value of the carbon center in adsorbed ethylene.



**Figure S2:** Ethylene deformation energy  $E_{\text{def}}(\text{C}_2\text{H}_4)$  via different modes as a function of mean hybridization value of the carbon center in adsorbed ethylene. The straight lines refer to linear fits.

## Reference

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