Epoxidation of Cardanol's Terminal Double Bond

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Supplementary Information

^{1}H-NMR analysis of SCECGE and TE-SCECGE resins.

Equations S1, S2, S3: Equations to calculate different monomer functionalities for SCECGE and TE-SCECGE resins.

Terminal double bond functionality: $$((I_B + I_D)/I_J) \times (3/3) \quad \text{(S1)}$$

Secondary epoxy functionality: $$(I_C)/I_J) \times (3/1) \quad \text{(S2)}$$

Primary epoxy functionality: $$(I_L)/I_J) \times (3/1) \quad \text{(S3)}$$

Table S1. Normalized peak intensity values as obtained via ^{1}H-NMR traces of the reactant SCECGE.

Peak Designation	Shift (ppm)	Peak Name	Normalized Intensity
B–D	5.2–5.8	Terminal double bond (-C**H**2=C**H**)	0.96 (3H)
C′	2.9–3.2	Secondary epoxy (CH-O-C**H**)	1.40 (1H)
L	3.4	Primary epoxy & terminal epoxy CH_2–C**H**-O-CH_2	1.05 (1H)
J	0.88	Terminal methyl (-C**H**3)	3 (3H)

Table S2. Normalized peak intensity values as obtained via ^{1}H-NMR traces of the product TE-SCECGE.

Peak Designation	Shift (ppm)	Peak Name	Normalized Intensity
B–D	5.2–5.8	Terminal double bond (-C**H**2=C**H**)	0.30 (3H)
C′	2.9–3.2	Secondary epoxy (CH-O-C**H**)	1.40 (1H)
L	3.4	Primary epoxy & terminal epoxy CH_2–C**H**-O-CH_2	1.25 (1H)
J	0.88	Terminal methyl (-C**H**3)	3 (3H)