Moisture Absorption and Opacity of Starch-Based Biocomposites Reinforced with Cellulose Fiber from Bengkoang

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Abstract: Cellulose fiber was isolated from bengkoang (\textit{Pachyrhizus erosus}) tuber peel. A suspension consisting of distilled water, starch, and glycerol was mixed with various cellulose loadings (0, 2, 6, and 10 g) then gelatinized using a hot plate with a magnetic stirrer. The biocomposite gel was sonicated using an ultrasonication probe (47.78 W/cm\textsuperscript{2} for 4 min). Scanning electron microscopy (SEM) micrographs for the fracture surface of resulting biocomposite films displayed a rougher surface than starch film, indicating fiber dispersion in the matrix. The opacity and moisture resistance of biocomposite films increased with the addition of cellulose. The opacity was at a maximum value (243.05 AU\textmu m) with 10 g fiber, which was 11.27\% higher than the starch film without cellulose. Moisture absorption of this biocomposite was 16.79\% lower than the starch film. Fourier transform infrared (FTIR) confirmed this more hydrophobic nature with lower transmittance at –OH stretching in the composite than the starch film. The addition of cellulose fiber into the matrix also increased the crystallinity index.

Keywords: biocomposite; starch; cellulose; ultrasonication; moisture absorption; opacity

1. Introduction

In recent years, the demand for the development of starch-based films has been high because of their biodegradability. Films produced from starch have applications in food packaging and agricultural and medical technologies. Starch is abundant and cheap, and provides good film formability [1]. However starch-based films have high water sensitivity, are brittle, and exhibit poor mechanical properties [2,3].

Natural fiber can be used to improve the properties of starch film. Research has shown that fibers from natural sources can potentially provide promising reinforcing materials. Natural fibers are easy to obtain, low cost, non-toxic, and have a high specific strength due to their low density [4,5]. In addition, natural fibers are environmentally friendly and require little energy in the composite fabrication process [6]. Many researchers have reported the improvement of starch-based biocomposites reinforced with natural fiber. They increase their oxygen barrier properties [7,8], water and moisture resistance [2,9,10], and mechanical properties [11–13].

Starch and biocomposite film properties also can be improved with ultrasonication. During the gelatinization process, ultrasonication helps to dissolve ghost particles and clumps of insoluble starch that weaken the resulting film’s mechanical properties [14,15]. Ultrasonication aids with dispersion...
and break-up of agglomerations of fibers in the matrix [16]. Also, ultrasonication is used in the production of nano-sized cellulose molecules because it produces high shear energy that breaks up long cellulose into shorter segments [17–19]. A reduction of cellulose size increases the area of contact surface between reinforcement and matrix, which improves the mechanical and barrier properties of the composite [20].

In last decade, the biocomposite potential of fibers and starches from various sources have been explored. Cassava, corn, sago, and potato starch have been used in the matrix. Fibers from potato tuber cells [10], banana peels [21,22], mandarin [16], grapefruit [23], and pomelo [24] have been used as reinforcement. Many of these starch and fiber sources can be obtained from agricultural products and waste. Another agricultural crop that produces both starch and cellulose fiber is bengkoang (Pachyrhizus erosus). This plant grows abundantly in tropical and subtropical regions, including in Indonesia, Mexico, Philippines, China, Malaysia, and Singapore. Information about starch extracted from the bengkoang tuber is still limited, and there has been no work reported on the use of the cellulose from bengkoang tuber peel as a reinforcement material. In this present study, we investigated the effect of various cellulose fiber loadings from bengkoang tuber peel on the properties of a starch-based biocomposite. Particle size distribution, opacity, SEM, X-ray diffraction (XRD), FTIR, and moisture absorption of the starch and biocomposite film were characterized.

2. Materials and Methods

2.1. Materials

Native bengkoang was obtained from a local farm at Padang, West Sumatra, Indonesia, and the starch (13.27% amylose content) was extracted. Glycerol purchased from Brataco (Jakarta, Indonesia) was used as a plasticizer. Distilled water was used as the plasticizing agent.

2.2. Isolation of Cellulose Fiber from Bengkoang Tuber Peel

Cellulose from bengkoang tuber peel was isolated using the method described by Julie Chandra et al. [17] with following modifications. Fibers were cut into 6 mm lengths. Alkaline treatment was used to remove hemicellulose and lignin content with 5% sodium hydroxide (NaOH) solution, and the suspension was heated at 50 °C for 4 h. After that, the alkali was washed from the fibers, and they were treated with 3.5 M hydrochloric acid (HCl) to break up the amorphous region, and the microfibrils were aggregate at 50 °C for 34 h. Treated fibers were washed with distilled water several times until pH = 7, then ground to a pulp manually with a mortar and pestle. This pulp was treated again with 5% NaOH solution at 50 °C for 4 h to remove the remaining non-cellulose content, and then subjected to acid hydrolysis with 5 M hydrochloric acid (HCl) at 50 °C for 17 h. At the end of process, a 4:1 mixture of sodium chlorite (NaClO2) and glacial acetic acid was used to bleach the fibers for 2 h at 60 °C. All chemical treatments were conducted using a hot plate magnetic stirrer with continuous stirring at 500 rpm. Distilled water was used to neutralize the fibers from acid. About 10 mL of bleached fiber was mixed with 140 mL distilled water and ultrasonicated with a 20 KHz ultrasonic cell crusher (Model SJIA-1200W, Ningbo Yinzhou Sjia Lab Equipment Co., Ltd., Ningbo, China) at 47.78 W/cm² for 120 min, the temperature being maintained under 40 °C.

2.3. Biocomposite Film Preparation

10 wt% starch, 4 g glycerol, and 100 mL distilled water were mixed in a 250 mL beaker (diameter 70 mm, IWAKI, Sumedang, Indonesia). A magnetic stirrer (Scilogex MS-H280-Pro, Scilogex LLC, Rocky Hill, CT, USA) was used at 500 rpm, while the suspension was heated to 80 °C. After the suspension was gelatinized, it was ultrasonicated for 4 min using a 20 KHz ultrasonic cell crusher (Model SJIA-1200W, Ningbo Yinzhou Sjia Lab Equipment Co., Ltd., Ningbo, China) at 47.78 W/cm². The suspension was then cast to a petri dish (15 cm diameter) and dried in a ventilated oven for 20 h at 50 °C. The resulting film was labelled as the starch film. Cellulose was added to the suspension at the
beginning of the similar processes, to fabricate composite films with different amounts of fiber loading (2, 6, and 10 g, equivalent to a dried cellulose weight of 0.16, 0.48, and 0.80 wt% respectively) and labelled as biocomposites (BC), followed by a numerical code indication of the amount of fiber solution added. All suspensions of biocomposite were poured into petri dishes and dried in a ventilated oven for 20 h at 50 °C. Table 1 shows the composition of starch and biocomposite films.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Starch (g)</th>
<th>Glycerol (g)</th>
<th>Distilled Water (g)</th>
<th>Cellulose Suspension (g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>10</td>
<td>4</td>
<td>100</td>
<td>-</td>
</tr>
<tr>
<td>BC-2</td>
<td>10</td>
<td>4</td>
<td>98</td>
<td>2</td>
</tr>
<tr>
<td>BC-6</td>
<td>10</td>
<td>4</td>
<td>94</td>
<td>6</td>
</tr>
<tr>
<td>BC-10</td>
<td>10</td>
<td>4</td>
<td>90</td>
<td>10</td>
</tr>
</tbody>
</table>

2.4. Characterization

2.4.1. Particle Size Distribution

Particle size measurement using dynamic light scattering (Zetasizer ZS, Malvern, UK) was used for the determination of particle size distribution of bengkoang tuber peel fiber. A 173° detector angle with a HeNe 4 mW 633 nm laser was used. Particle size determination was repeated three times.

2.4.2. Film Opacity

Opacity of films was determined with a spectrophotometer (Shimadzu UV 1800, Shimadzu Corporation, Kyoto, Japan). Films were cut into 1 cm × 2.5 cm rectangles and fixed onto the inner side of a spectrophotometer cell. The absorbance spectrum between 400 and 800 nm was recorded. The opacity of films was determined as the area under spectrum according to ASTM D 1003-00 (Standard test method for haze and luminous transmittance of transparent plastics). The opacity determinations were repeated three times.

2.4.3. X-ray Diffraction (XRD)

A PANalyticalXpert Pro diffractometer with Cu Kα radiation at 40kV and 30 mA was used to record XRD diffractograms of films. The diffraction between 2θ = 10° to 40° was scanned. The crystallinity index (CI) of films was determined by calculating the ratio of crystalline area to the total area on XRD diffractograms [8]. CI for fiber was calculated with following equation [25]:

\[
CI = \frac{I_{cry} - I_{am}}{I_{cry}}
\]

where \(I_{cry}\) was the maximum intensity of crystalline peak at 2θ = 22–24°, and \(I_{am}\) was the minimum intensity of amorphous diffraction at 2θ = 18–20°.

2.4.4. Fourier Transform Infrared Spectroscopy

FTIR spectra of films were recorded with a FTIR spectrometer (Frontier, PerkinElmer, Waltham, MA, USA) within the wavenumber range of 4000–600 cm\(^{-1}\) at 4 cm\(^{-1}\) resolution.

2.4.5. Moisture Absorption

Moisture absorption of all films was determined using the method described by Abral et al. [9] with the following modifications. Films were cut to 2 cm × 1 cm size and dried in a ventilated oven at 50 °C for 24 h. Dried pieces were weighed to determine the initial mass, then they were placed in a covered box containing saturated NaCl solution (Relative Humidity (RH) 75%) at 25 °C ± 2 °C. Every 30 min, all pieces were weighed.
2.4.6. Scanning Electron Microscopy

The morphology of starch and biocomposite film fracture surfaces were analyzed using a scanning electron microscope (HITACHI SU-3500, Hitachi High-Technologies Corporation, Tokyo, Japan) at an accelerating voltage of 5 kV in a $5 \times 10^{-4}$ Pa vacuum. All samples were coated with gold (Au).

3. Results and Discussion

3.1. Particle Size Distribution

Figure 1 shows the size distribution of the cellulose particles produced from bengkoang peel fiber. 24.3% of the cellulose particles were below 100 nm in length which is within the nano-size range, but the other 72.5% were in the 100–250 nm micro-sized range. The reduction of cellulose size started from the removal of the amorphous components from the cellulose after acid hydrolysis [26,27], and then it was reduced further with ultrasonication as the high shear forces and cavitation energy broke up the particles. It also broke apart the aggregates of cellulose, enabling for better dispersion in the aqueous suspension. Similar results have also been reported by Abral et al. [18], who isolated nanocellulose from bacterial cellulose, and Niu et al. [28], who isolated nanocellulose from microcrystalline cellulose.

![Figure 1. Particle size distribution of cellulose fiber from bengkoang tuber peel.](image)

3.2. Film Opacity

Figure 2 shows the opacity of starch and biocomposite films. The addition of cellulose in the matrix increased the transparency of the biocomposites. For example, the opacity of the starch film was 218.44 AUmm, which was lower than that of BC-10 (243.05 AUmm, 11.27% higher). This is because cellulose is more opaque than starch film. Previous studies also reported on the increasing opacity of biocomposites with increasing fiber content [2,22].
3.3. X-ray Diffraction

Figure 3 shows the XRD patterns of fibers, starch films, and biocomposite films. As shown in Figure 3b, two major peaks were observed at around $2\theta = 17^\circ$ and $22.2^\circ$. All films showed similar curves, but the diffraction intensity increased as the amount of cellulose increased from 1211 (Control) to 1351 (BC-10) at around $2\theta = 22.2^\circ$. This result is because the presence of cellulose in the matrix CI for fiber was 58.48% higher than the starch and biocomposite films. In this case, crystallinity index (CI) for BC-10 was 2.51% higher than that for the control. A similar result was also reported by Abral et al. [13]; they reported the increasing CI of the composites with the addition of fibers.
3.4. Fourier Transform Infrared

Figure 4 displays the FTIR spectra of starch and biocomposite films. All FTIR spectra showed similar patterns. Ultrasonication did not result in the appearance of new functional groups, but it shifted the transmittance intensity of the pattern. The band, at around 3296 cm$^{-1}$ corresponded to the complex vibrational stretching of hydrogen bonded hydroxyl groups (–OH) [29]. Other bands were visible around 2926 cm$^{-1}$ (–CH stretching), 1647 cm$^{-1}$ (H–O–H stretching, absorbed water), 1353 cm$^{-1}$ (–CO stretching in an aryl-aryl ether) and 1013 (–CO stretching in an anhydroglucose ring). Transmittance intensity of all bands increased with an increase in fiber loading.

FTIR peaks can be also used to determine the change in the crystalline structure of films. This is assessed as the absorbance ratio of the peaks at 995/1022, this being the ratio between the crystalline and the amorphous regions of the biopolymer [14]. As shown in Table 2, the ratio was increased, as
the fiber loading increased. For example, the ratio for BC-10 (0.59) was 22.92% higher than BC-2 (0.48). This is consistent with the crystallinity index calculated from the XRD pattern.

**Table 2.** Crystallinity index of starch and biocomposite films.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Crystallinity Index (%)</th>
<th>Intensity Ratio at 1022/995</th>
</tr>
</thead>
<tbody>
<tr>
<td>Fiber</td>
<td>58.48</td>
<td>-</td>
</tr>
<tr>
<td>Control</td>
<td>44.70</td>
<td>0.47</td>
</tr>
<tr>
<td>BC-2</td>
<td>44.75</td>
<td>0.48</td>
</tr>
<tr>
<td>BC-6</td>
<td>45.63</td>
<td>0.54</td>
</tr>
<tr>
<td>BC-10</td>
<td>45.82</td>
<td>0.59</td>
</tr>
</tbody>
</table>

3.5. Moisture Absorption

Figure 5 shows the moisture absorption of starch and biocomposite films. The moisture absorption of the starch film was higher than that of the biocomposites. This is because the starch film was more hydrophilic than the biocomposites as reflected in the lower transmittance of –OH stretching bands than in the biocomposite. Adding 2 g fiber decreased the moisture absorption after 5 h by 3.85%. With further addition of the cellulose fiber, the moisture absorption continued to decrease. For example, the moisture absorption of BC-10 was 13.46% lower than that of BC-2. This was because the number of accessible hydroxyl groups in BC-10 was less than in BC-2. The addition of cellulose fiber reduced the total available –OH groups because of increased intermolecular bonding. This is consistent with the FTIR pattern (Figure 4) which confirmed the more hydrophobic nature of biocomposites with higher cellulose fiber loading in the matrix. For example transmittance (T) of –OH stretching for BC-2 was 2.20% higher than in the control. T value increased with increasing fiber loading, for instance BC-10 had a T value 29.75%, higher than that of BC-2. This result confirmed that BC-10 was more hydrophobic than BC-2. Moisture absorption of biocomposite film decreased as increasing fiber loading. Similar result have also been reported in a previous study which showed increasing moisture resistance of a biocomposite reinforced by higher fiber loading [13]. This is also consistent with the XRD curve (Figure 3). Similar agreement was also reported by Dufresne et al. [10], Montero et al. [20] and Abral et al. [30].

![Figure 5. Moisture absorption for control and biocomposite films for 5 h testing time.](image-url)
3.6. Scanning Electron Microscopy

Figure 6a–h show the SEM micrographs of the cross-sections of starch and biocomposite films. As can be seen in Figure 6c,d, the biocomposite film fracture surfaces were slightly rougher compared to the starch films (Figure 6a,b). Increases in cellulose suspension loading resulted in an increased roughness of fracture surfaces (Figure 6e–h). There is no agglomeration of cellulose fibers observed on the surface. This indicates that fibers were well dispersed in the starch matrix, and it may indicate increases in interfacial hydrogen bonding and homogeneity of the biocomposite structure. This result is in a good agreement with the previous study [13]. The higher compactness between the matrix and fiber led to an increase in the crystallinity index as shown in Table 2. The better interfacial hydrogen bonding reduced the number of accessible -OH groups, thus increasing moisture resistance of the biocomposites. This result is supported by the moisture absorption data in Figure 5, which shows a decrease in moisture absorption with increased fiber loading.

![Figure 6](image-url)
Figure 6. SEM micrograph of fractured cross section of the (a) Control; (b) BC-2; (c) BC-6; (d) BC-10 in X250; (e) Control; (f) BC-2; (g) BC-6; (h) BC-10 in X1000.

4. Conclusions

This study investigated the properties of starch-based biocomposites reinforced with cellulose fiber isolated from bengkoang tuber peel. Overall, cellulose fiber improves the opacity and moisture absorption of the starch-based film and within the range of cellulose fiber concentrations tested (up to 0.80 wt% for each 10 wt% of starch), the higher the cellulose fiber, the higher the opacity and moisture resistance. While moisture resistance is obviously important for the resilience of biofilm packaging, opacity is also an important parameter where the biofilms are used to contain items that need to be protected from light. The addition of cellulose fibers also increased the crystallinity index of the films. A high crystallinity index, which is related to the close and regular packing of polymers and microfibers, is likely to be an indication of superior mechanical properties. This indicates that bengkoang tuber peel fibers that were purified and reduced to micro-size in this manner have potential as a reinforcement material in biocomposite applications.

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Conflicts of Interest: The authors declare no conflict of interest.

References


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